

Slides from 2012 in videos assigned
for lectures on 8 and 10 November

2012-Lecture 10 starting from 0:33
until the end, all of lecture 11 and
lecture 12 from start to 1:02.

INTENSITIES IN PHOTOELECTRON SPECTRA:

- GENERAL: FINAL STATE K (K-SUBHELL + ALL OTHER DESIG.)

$$\text{INT.}_K \propto |\hat{e} \cdot \langle \Psi_{\text{tot}}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_{\text{tot}}^i(N) \rangle|^2 \quad (\text{DIPOLE APPROX.})$$

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- BORN-OPPENHEIMER: e⁻'S FAST, VIBRATIONS SLOW

$$\text{INT.}_K \propto |\underbrace{\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2}_{\text{FRANCK-CONDON FACTOR}} | \hat{e} \cdot \langle \Psi_e^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle \rangle|^2$$

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- BORN-OPPENHEIMER: e⁻'S FAST, VIBRATIONS SLOW

$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\hat{e} \cdot \langle \Psi_{e^-}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_{e^-}^i(N) \rangle|^2$$

FRANCK-CONDON FACTOR

- SUDDEN APPROXIMATION: $\Psi_K \rightarrow \Psi_f = \text{PHOTOE}^-$ (FAST)



$$\text{INT.}_K \propto |\langle \Psi_{\text{VIB}, v}^f | \Psi_{\text{VIB}, v}^i \rangle|^2 |\underbrace{\langle \Psi_{e^-}^f(N-1, K) | \Psi_{e^-}^{i-}(N-1, K) \rangle}_{|\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_K \rangle|^2} |^2$$

SAME SUBSHELL COUPLING +
 \hookrightarrow NORMAL $\frac{dG_K}{d\omega}$ TOTAL L,S → "MONOPOLE"

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- SUDDEN APPROXIMATION: $\Psi_K \rightarrow \Psi_f = \text{PHOTOE}^-$ (FAST)



$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\underbrace{\langle \Psi_{e^-}^f(N-1, K) | \Psi_R^i(N-1, K) \rangle}_{{\color{red} |\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_L \rangle|^2 \text{ SAME SUBSHELL COUPLING + } \hookrightarrow \text{NORMAL } \frac{dG_K}{d\Omega} \text{ TOTAL L,S} \rightarrow \text{"MONPOLE"}}|^2$$

- SLATER DETS. FOR $\Psi_{e^-}^f = \det(\Psi'_1 \Psi'_2 \dots \Psi'_{K-1} \Psi'_{K+1} \dots \Psi'_N)$

$$\Psi_R = \det(\Psi_1 \Psi_2 \dots \Psi_{K-1} \Psi_{K+1} \dots \Psi_N)$$

$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\langle \Psi'_1 | \Psi'_1 \rangle|^2 |\langle \Psi'_2 | \Psi'_2 \rangle|^2 \dots |\langle \Psi'_{K-1} | \Psi'_{K-1} \rangle|^2 / |\langle \Psi'_{K+1} | \Psi'_{K+1} \rangle|^2 \dots |\langle \Psi'_N | \Psi'_N \rangle|^2$$

$$|\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_L \rangle|^2$$

1e- DIPOLE → $d\sigma/d\Omega$

**(N-1)e- SHAKE-UP/
SHAKE-OFF →
"MONPOLE"**

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- BORN-OPPENHEIMER: e-'s FAST, VIBRATIONS SLOW

$$\text{INT.}_K \propto |\underbrace{\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2}_{\text{FRANCK-CONDON FACTOR}} |\hat{e} \cdot \langle \Psi_{e^-}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_{e^-}^i(N) \rangle|^2$$

- SUDDEN APPROXIMATION: $\Psi_K \rightarrow \Psi_f = \text{PHOTOE}^-$ (FAST)



$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\underbrace{\langle \Psi_{e^-}^f(N-1, K) | \Psi_{e^-}^{i-}(N-1, K) \rangle}_{{\color{red} |\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_L \rangle|^2}}|^2$$

SAME SUBSHELL COUPLING +
 ↳ NORMAL $\frac{dG_K}{d\Omega}$ TOTAL L,S → "MONPOLE"

- SLATER DETS. FOR $\Psi_{e^-}^f = \det(\Psi'_1 \Psi'_2 \dots \Psi'_{K-1} \Psi'_{K+1} \dots \Psi'_N)$

$$\Psi_L = \det(\Psi_1 \Psi_2 \dots \Psi_{K-1} \Psi_{K+1} \dots \Psi_N)$$

$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\langle \Psi'_1 | \Psi_L \rangle|^2 |\langle \Psi'_2 | \Psi_L \rangle|^2 \dots |\langle \Psi'_{K-1} | \Psi_L \rangle|^2 / |\langle \Psi'_{K+1} | \Psi_L \rangle|^2 \dots |\langle \Psi'_N | \Psi_L \rangle|^2$$

$$|\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_L \rangle|^2$$

1e- DIPOLE → $d\sigma/d\Omega$

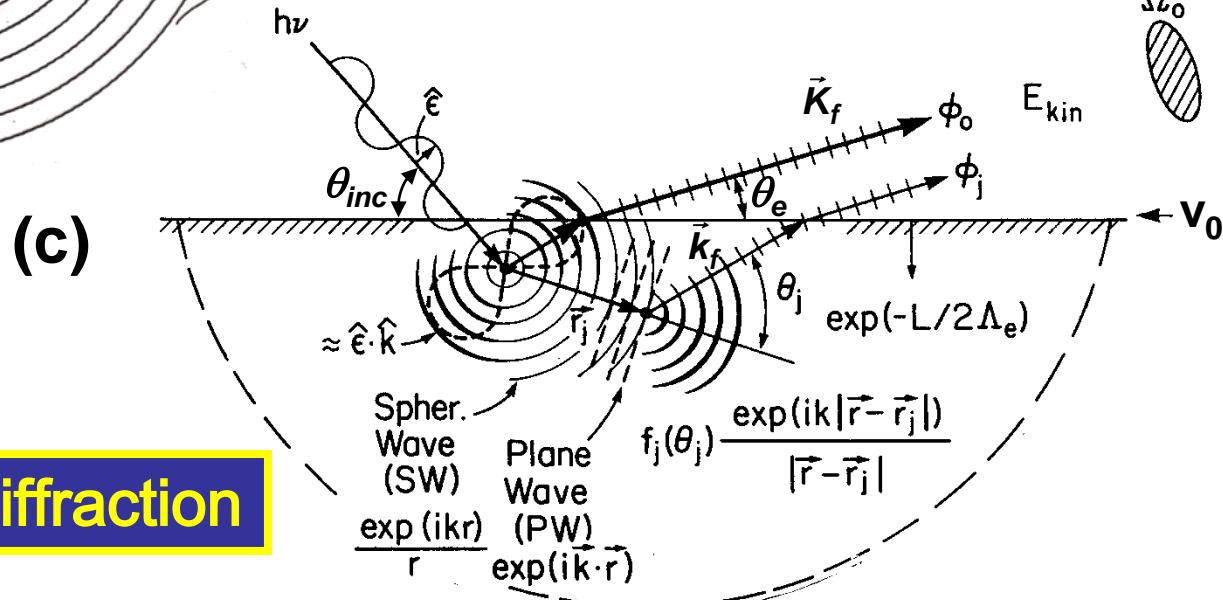
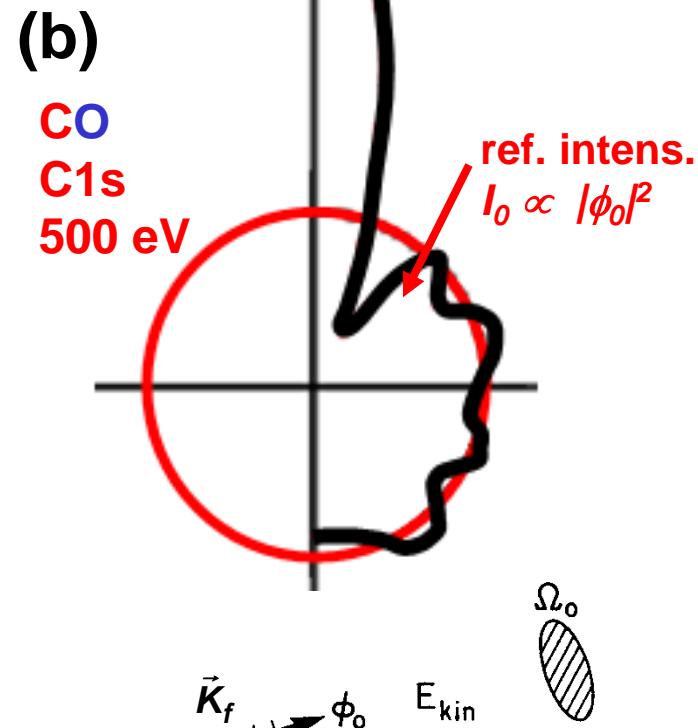
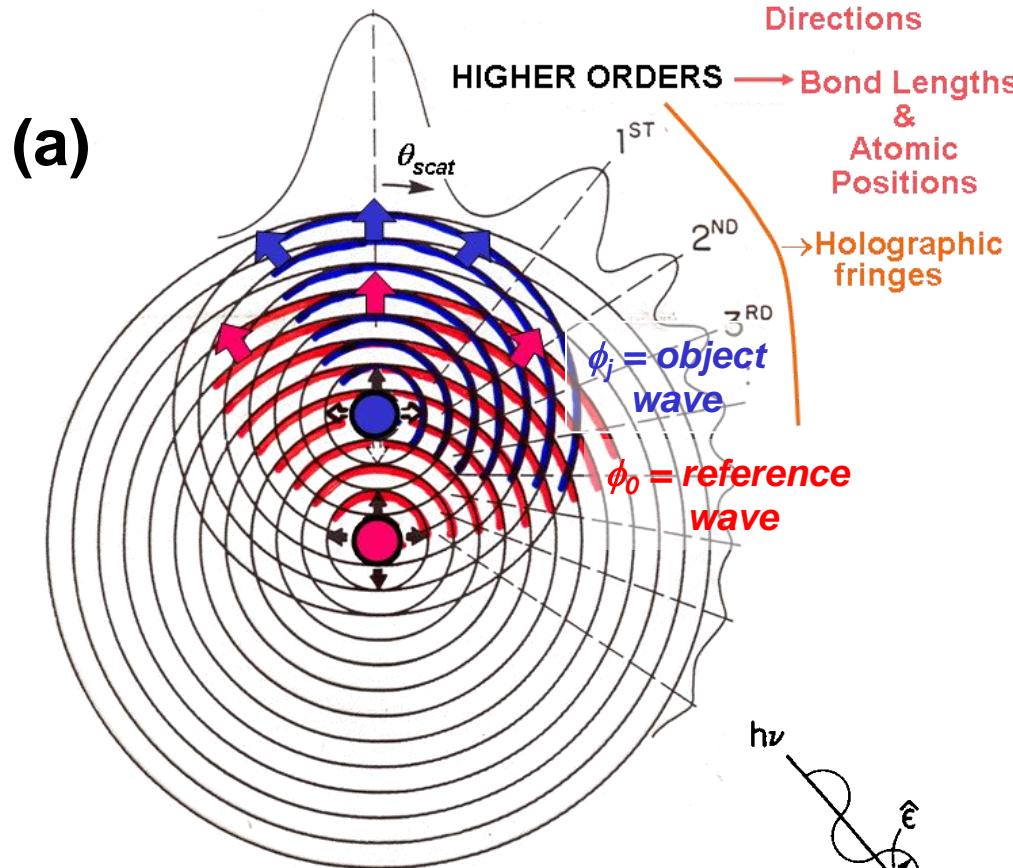
**(N-1)e- SHAKE-UP/
SHAKE-OFF →
"MONPOLE"**

"Basic Concepts of XPS"
Section 3.D.

- PLUS DIFFRACTION EFFECTS IN Ψ_f ESCAPE

Section 6.D.

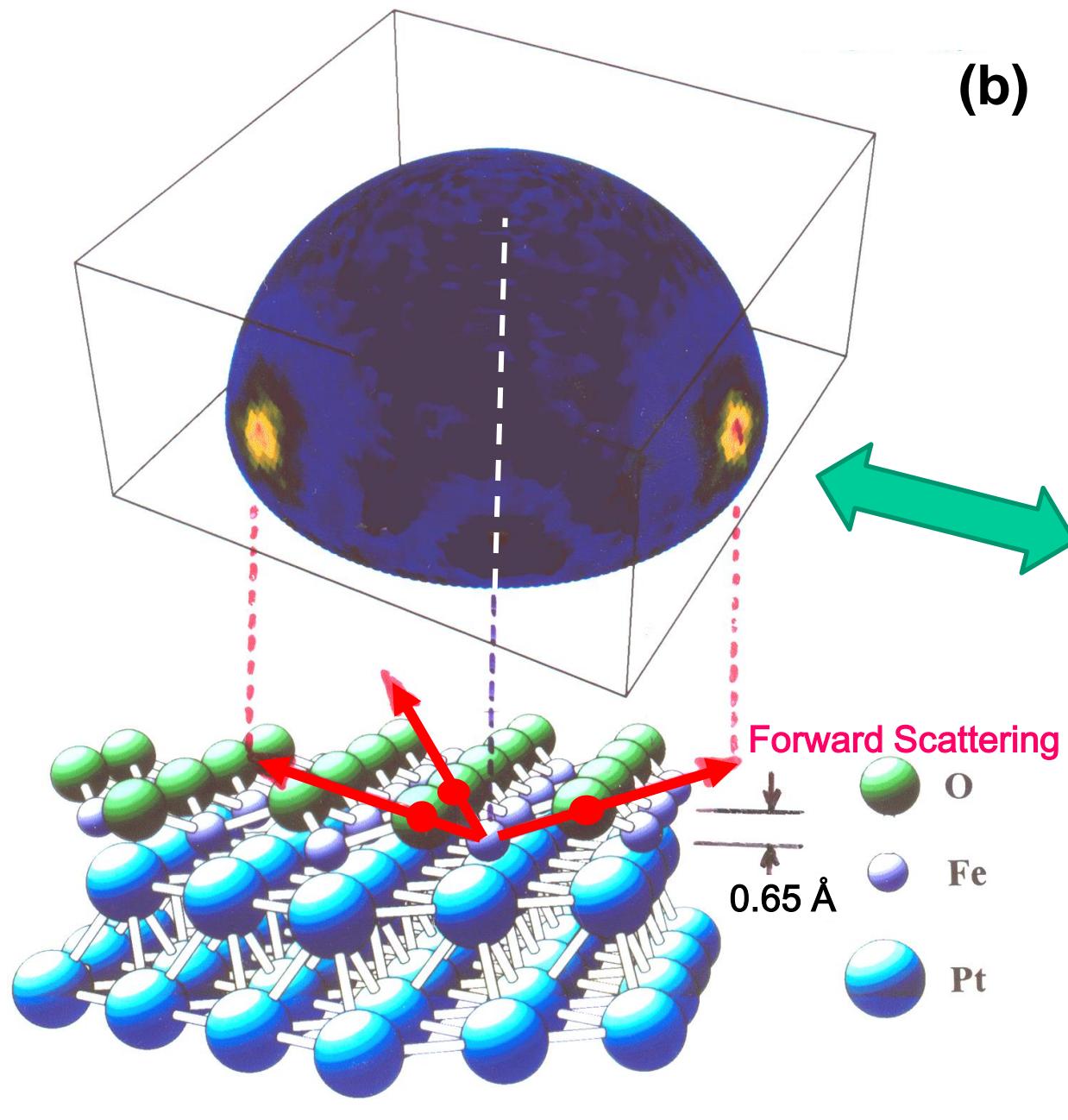
FORWARD SCATT. = "0TH ORDER" → Bond & Low-Index Directions



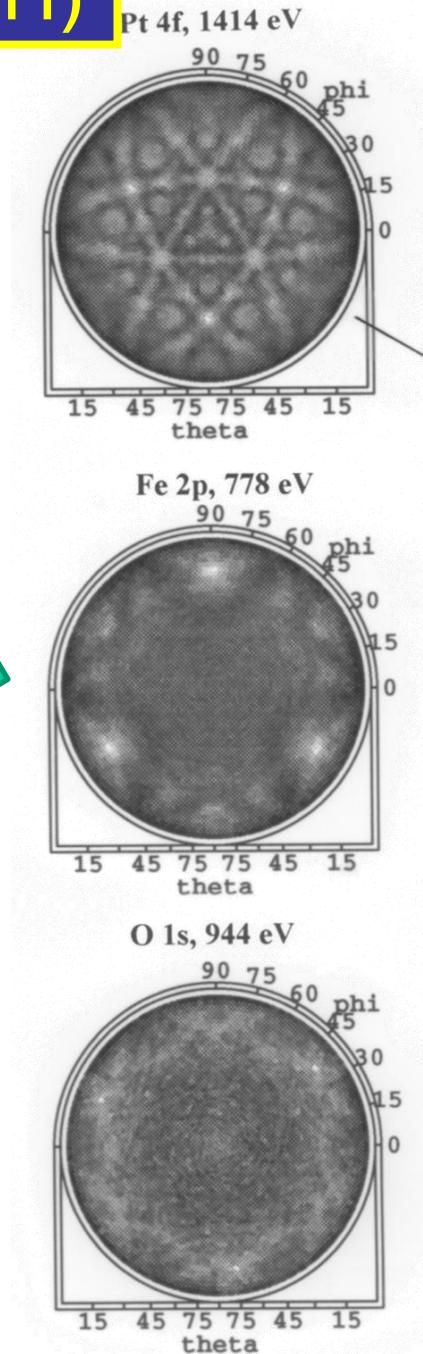
Photoelectron Diffraction

X-ray Photoelectron Diffraction: 1ML FeO on Pt(111)

(a)



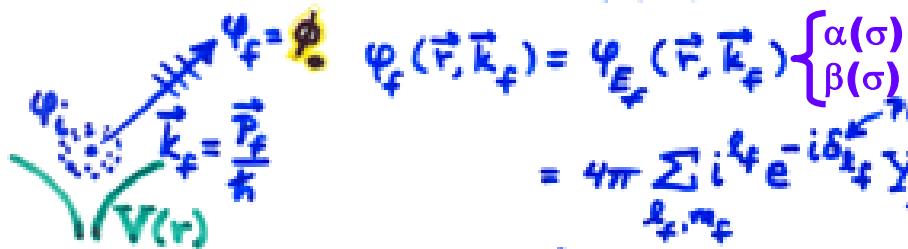
(b)



PHOTOELECTRON EMISSION - BASIC MATRIX ELEMENTS + SELECTION RULES :

- ATOMIC-LIKE (LOCALIZED) STATES \Rightarrow CORE:

$$\Psi_i(\vec{r}) = \Psi_{n_i l_i m_i}(r, \theta, \phi) = R_{n_i l_i}(r) Y_{l_i m_i}(\theta, \phi) \quad \left\{ \begin{array}{l} \alpha(\sigma) = m_{si} = +\frac{1}{2} = \uparrow \\ \beta(\sigma) = m_{si} = -\frac{1}{2} = \downarrow \end{array} \right.$$



$$\Psi_f(\vec{r}, \vec{k}_f) = \Psi_{E_f}(\vec{r}, \vec{k}_f) \quad \left\{ \begin{array}{l} \alpha(\sigma) \\ \beta(\sigma) \end{array} \right.$$

$$= 4\pi \sum_{l_f m_f} i^{l_f} e^{-i\delta k_f} Y_{l_f m_f}^*(\theta, \phi) Y_{l_f m_f}(\theta, \phi) R_{E_f l_f}(r) \quad \left\{ \begin{array}{l} \alpha(\sigma) \\ \beta(\sigma) \end{array} \right.$$

PHASE SHIFT OF k_f WAVE IN $V(r)$

DIPOLE APPROX.: INT. $\propto |\langle \Psi_f | \hat{\mathbf{e}} \cdot \vec{r} | \Psi_i \rangle|^2 = |\langle \Psi_f | \vec{r} | \Psi_i \rangle|^2 \Rightarrow \langle \Delta l = l_f - l_i = \pm 1$

EQUIVALENT
WITHIN CONSTANT
FACTOR



TWO CHANNELS
 $\Delta m = m_f - m_i = 0, \pm 1$
 LINEAR POLARIZ.
 $\Delta m = \pm 1$, CIRCULAR POLARIZATION

$$\Delta m_s = m_{sf} - m_{si} = 0 !$$

The quantum mechanics of covalent bonding in molecules: H_2^+ with one electron

$$\varphi_- = \varphi_{\text{antibonding}} \\ \approx \varphi_{1sa} - \varphi_{1sb}$$

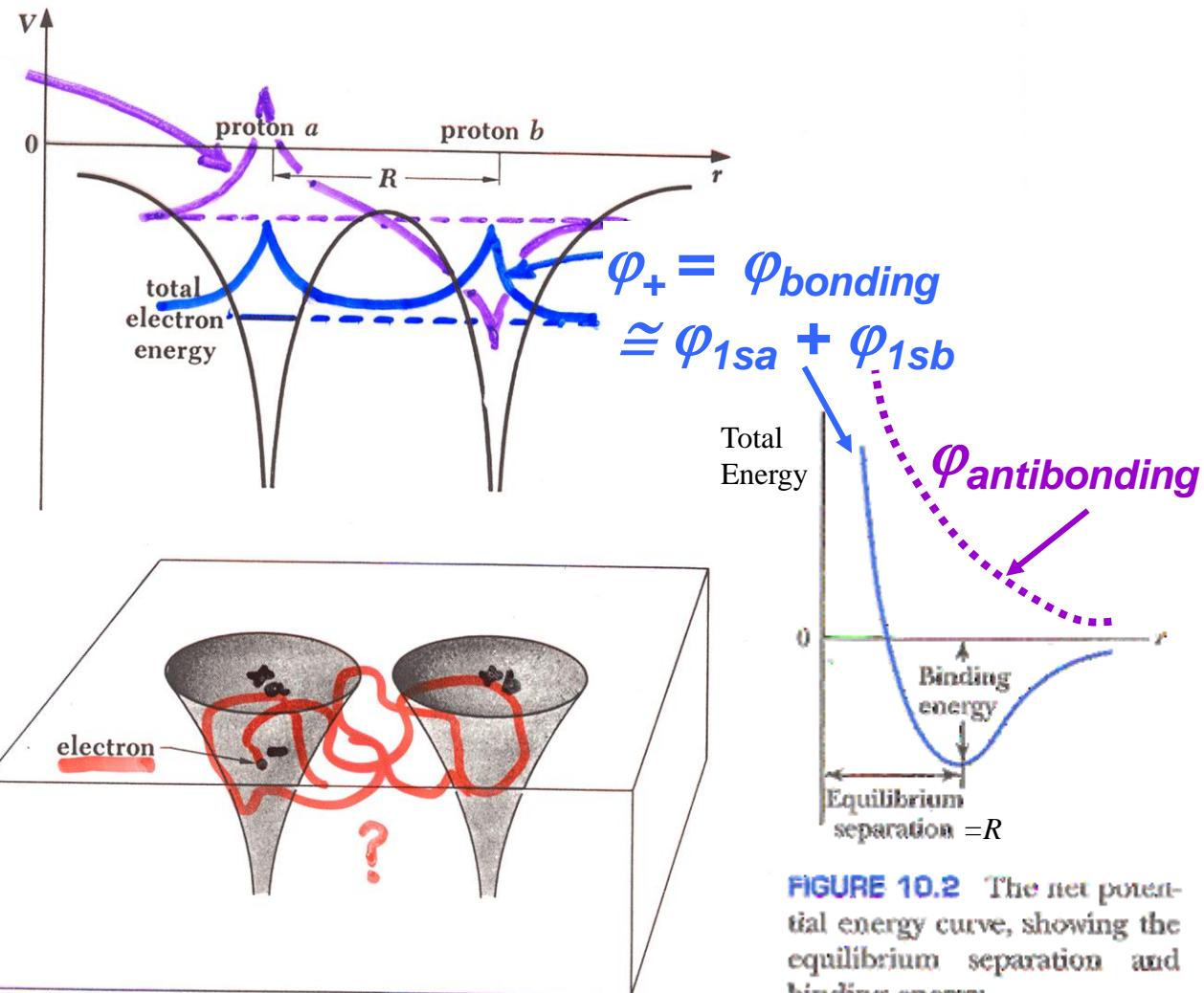
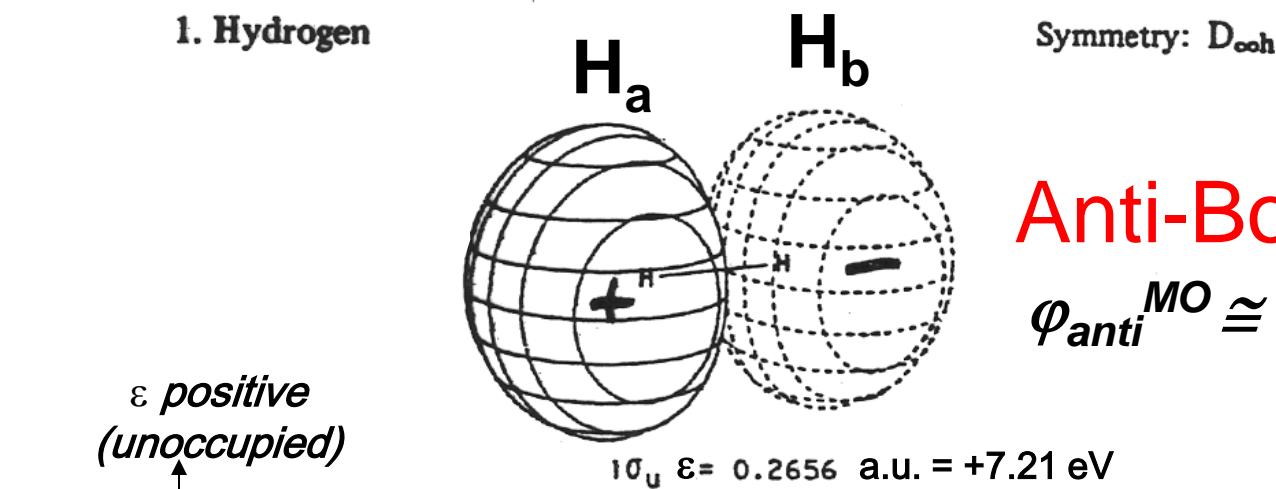


FIGURE 8.4 (a) Potential energy of an electron in the electric field of two nearby protons. The total energy of a ground-state electron in the hydrogen atom is indicated. (b) Two nearby protons correspond quantum-mechanically to a pair of boxes separated by a barrier.

FIGURE 10.2 The net potential energy curve, showing the equilibrium separation and binding energy.

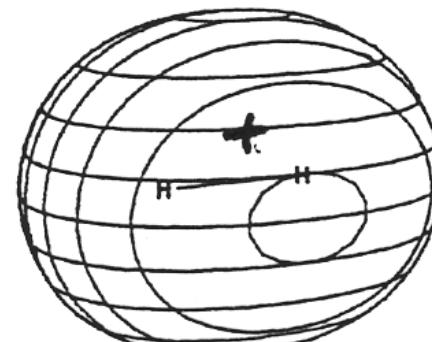
1. Hydrogen



Anti-Bonding

$$\varphi_{\text{anti}}^{\text{MO}} \cong \varphi_{1\text{sa}} - \varphi_{1\text{sa}}$$

ϵ negative
(occupied)



$$1\sigma_g \quad \epsilon = -0.5944 \text{ a.u.} = -16.16 \text{ eV}$$

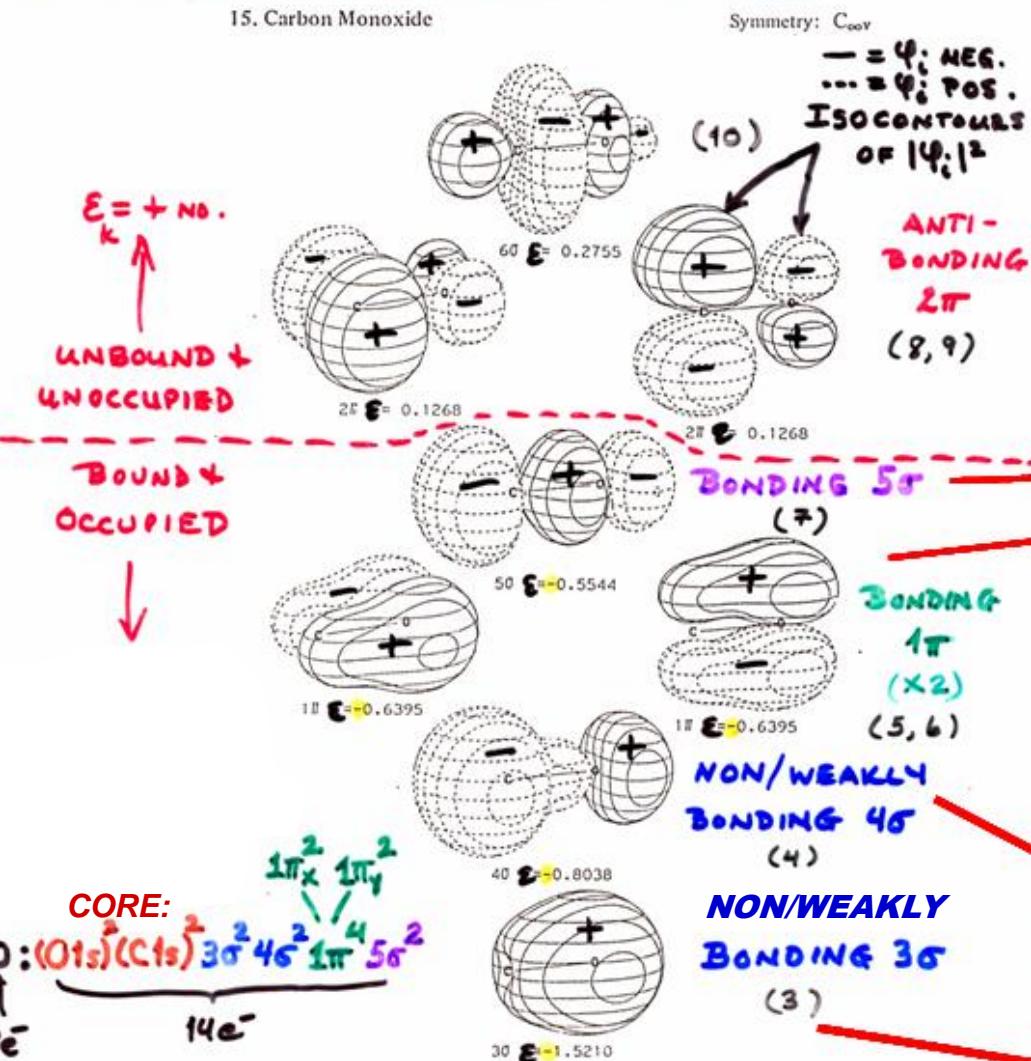
(Compare - 13.61 for H atom 1s)

Bonding

$$\varphi_{\text{bonding}}^{\text{MO}} \cong \varphi_{1\text{sa}} + \varphi_{1\text{sa}}$$

The LCAO or tight-binding picture for CO:

15. Carbon Monoxide



Atomic orbital makeup

$$\varphi_j^{\text{MO}}(\vec{r}) = \sum_{\text{Atoms A}} \sum_{\text{Orbitals } i} c_{Ai,j} \varphi_{Ai}^{\text{AO}}(\vec{r})$$

Atoms A
Orbitals i

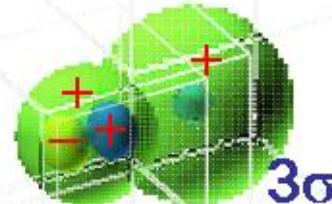
5σ



$1\pi_x$

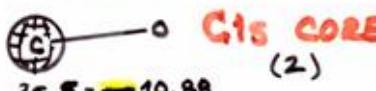


4σ



3σ

Chemist's picture (no core):



$2\sigma E = -10.88$



$1\sigma E = -20.00 \text{ a.u. } (1 \text{ a.u.} = 21.2 \text{ eV})$



THE ELECTRONS IN HF (OR HCl): ionic molecules

Symmetry: $C_{\infty v}$

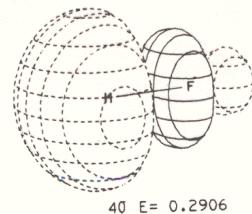
10. Hydrogen Fluoride = HF

MO's LIKE

HCl

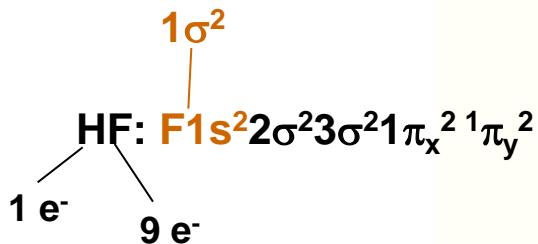
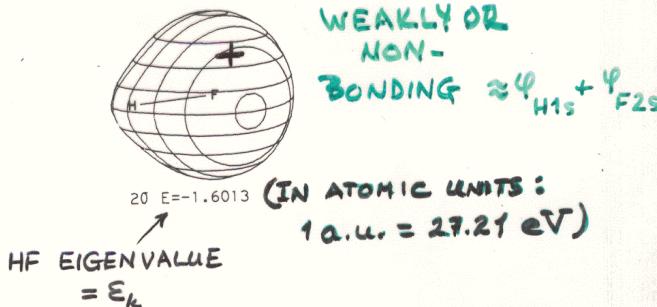
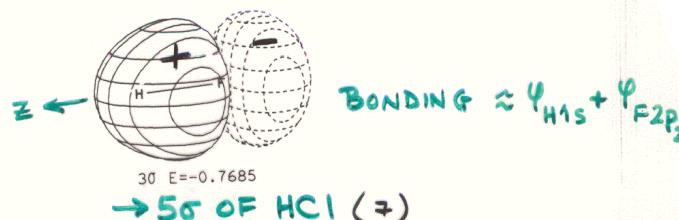
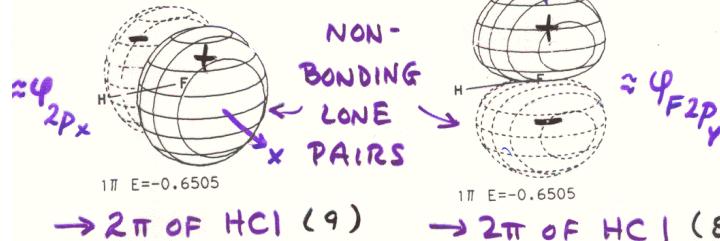
$2s, 2p \rightarrow$

$3s, 3p$



ANTI-BONDING
(UNOCCUPIED)

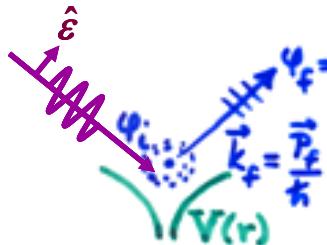
UNOCC.
OCC.



PHOTOELECTRON EMISSION - BASIC MATRIX ELEMENTS + SELECTION RULES:

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$$\Psi_f(\vec{r}, \vec{k}_f) = \Psi_{E_f}(\vec{r}, \vec{k}_f) \quad \left\{ \begin{array}{l} \alpha(\sigma) \\ \beta(\sigma) \end{array} \right.$$

= $4\pi \sum_{l_f, m_f} i^{l_f} e^{-i\delta_{k_f}} Y_{l_f m_f}^*(\theta, \phi) Y_{l_f m_f}(\theta, \phi) R_{E_f}(r) \quad \left\{ \begin{array}{l} \alpha(\sigma) \\ \beta(\sigma) \end{array} \right.$

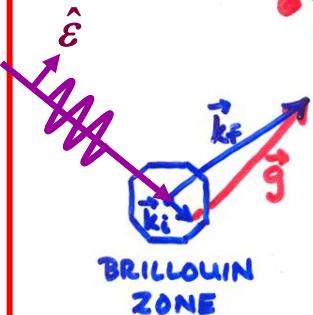
DIPOLE APPROX.: INT. $\propto |\langle \Psi_f | \hat{E} \cdot \vec{r} | \Psi_i \rangle|^2 = |\hat{E} \cdot \langle \Psi_f | \vec{r} | \Psi_i \rangle|^2 \Rightarrow \Delta l = l_f - l_i = \pm 1$

EQUIVALENT WITHIN CONSTANT FACTOR $\left(\begin{array}{c} \downarrow \\ \vec{p} \\ \uparrow \\ \vec{\nabla} V(r) \end{array} \right)$

$\Delta m = m_f - m_i = 0, \pm 1$
 LINEAR POLARIZ.
 $\Delta m = \pm 1$, CIRCULAR POLARIZATION

$\Delta m_s = m_{sf} - m_{si} = 0 !$

- BLOCH-FUNCTION (DELOCALIZED) STATES \Rightarrow VALENCE:



$$\Psi_i(\vec{r}) = U_{\vec{k}_i}(\vec{r}) e^{i\vec{k}_i \cdot \vec{r}}$$

$$\Psi_f(\vec{r}) = U_{\vec{k}_f}(\vec{r}) e^{i\vec{k}_f \cdot \vec{r}} ; E_f = \frac{p_f^2}{2m} = \frac{\hbar^2 k_f^2}{2m} \quad \text{USUALLY NEGIG.}$$

$$|\langle \Psi_f | \hat{E} \cdot \vec{r} | \Psi_i \rangle|^2 = |\hat{E} \cdot \langle \Psi_f | \vec{r} | \Psi_i \rangle|^2 \Rightarrow \Delta \vec{k} = \vec{k}_f - \vec{k}_i - \vec{k}_{hv} + \vec{k}_{\text{PHON-ON}} \\ = \vec{g}_{\text{BULK}} \text{ (or } \vec{g}_{\text{SURF}} \text{)}$$

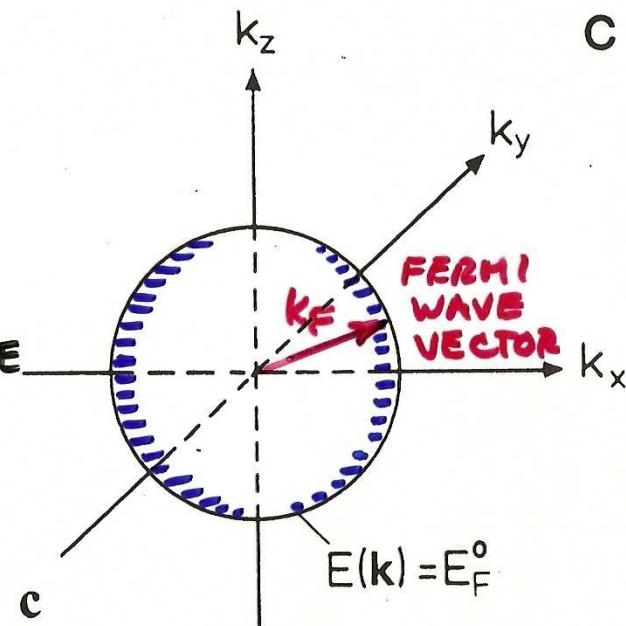
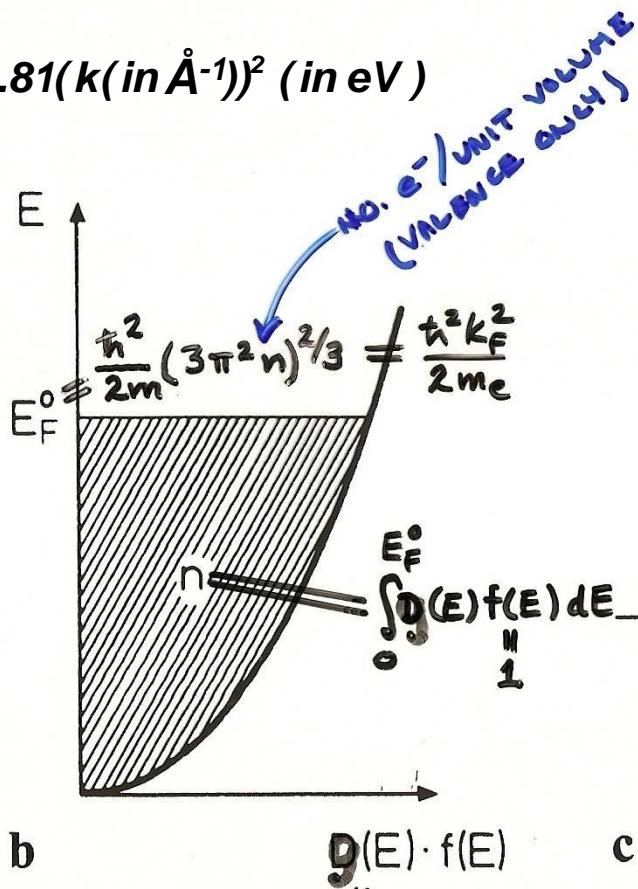
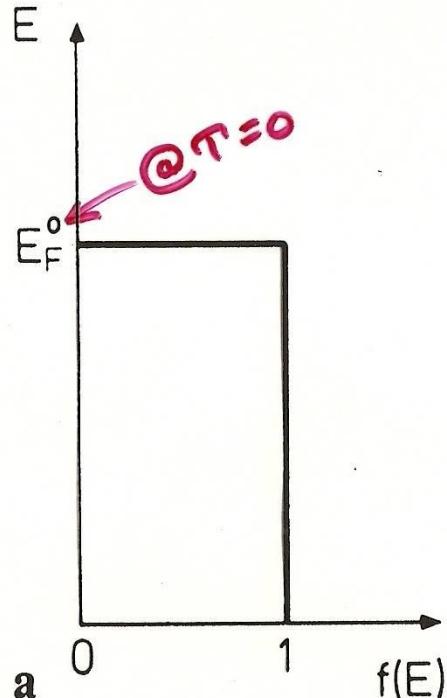
"DIRECT" TRANSITIONS

BUT LATTICE VIBRATIONS \Rightarrow SUM OVER \vec{k}_{PHONON}

\Rightarrow FRACTION DIRECT \approx DEBYE-WALLER FACTOR
 $= \exp[-g^2 \bar{u}^2]$

The free-electron solid at absolute zero

$$E(\vec{k}) = \frac{\hbar^2 k^2}{2m_e} = 3.81(k \text{ (in } \text{\AA}^{-1}))^2 \text{ (in eV)}$$



FERMI-DIRAC

$$f(E,T) = \frac{1}{e^{(E-E_F^0)/k_B T} + 1}$$

= the density of states

$$\frac{(2m)^{3/2}}{2\pi^2 \hbar^3} E^{1/2}$$

$$k_F = (3\pi^2 n)^{1/3}$$

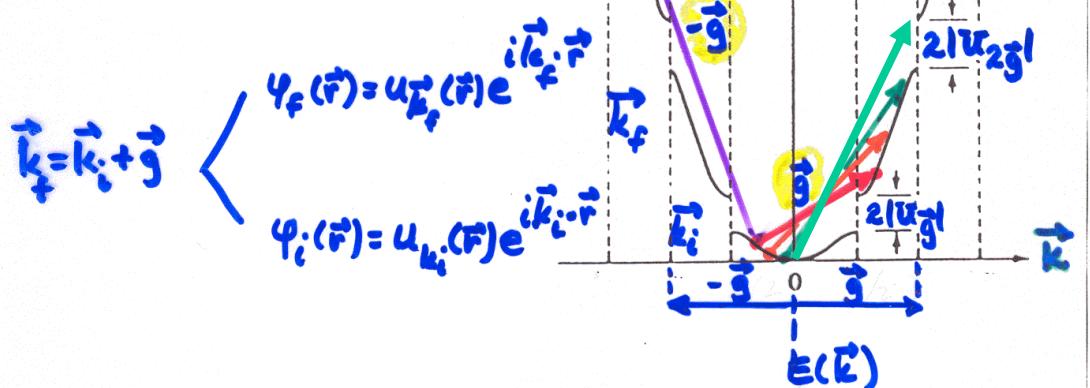
NEARLY-FREE ELECTRONS IN A WEAK PERIODIC POTENTIAL—1 DIM.

$E(\vec{k})$

EXTENDED ZONE:

$$V(\vec{r}) = \sum_{\vec{g}} U_{\vec{g}} e^{i\vec{g} \cdot \vec{r}}$$

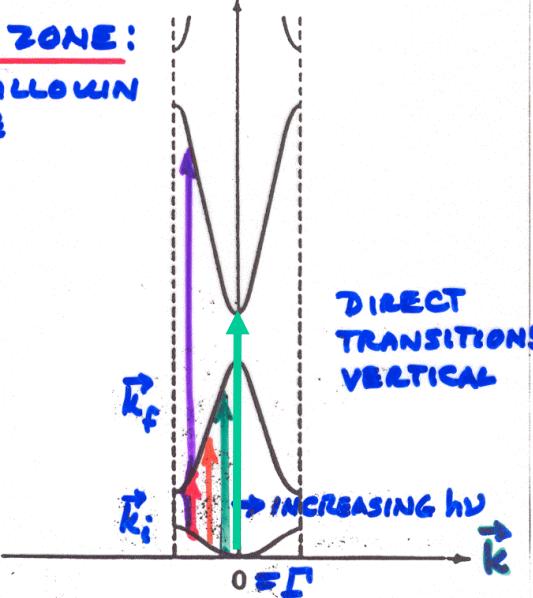
$$E(\vec{k}) \approx \frac{\hbar^2 k^2}{2m} + \text{gaps}$$



$E(\vec{k})$

REDUCED ZONE:
= FIRST BRILLOUIN ZONE

DIRECT
TRANSITIONS
VERTICAL



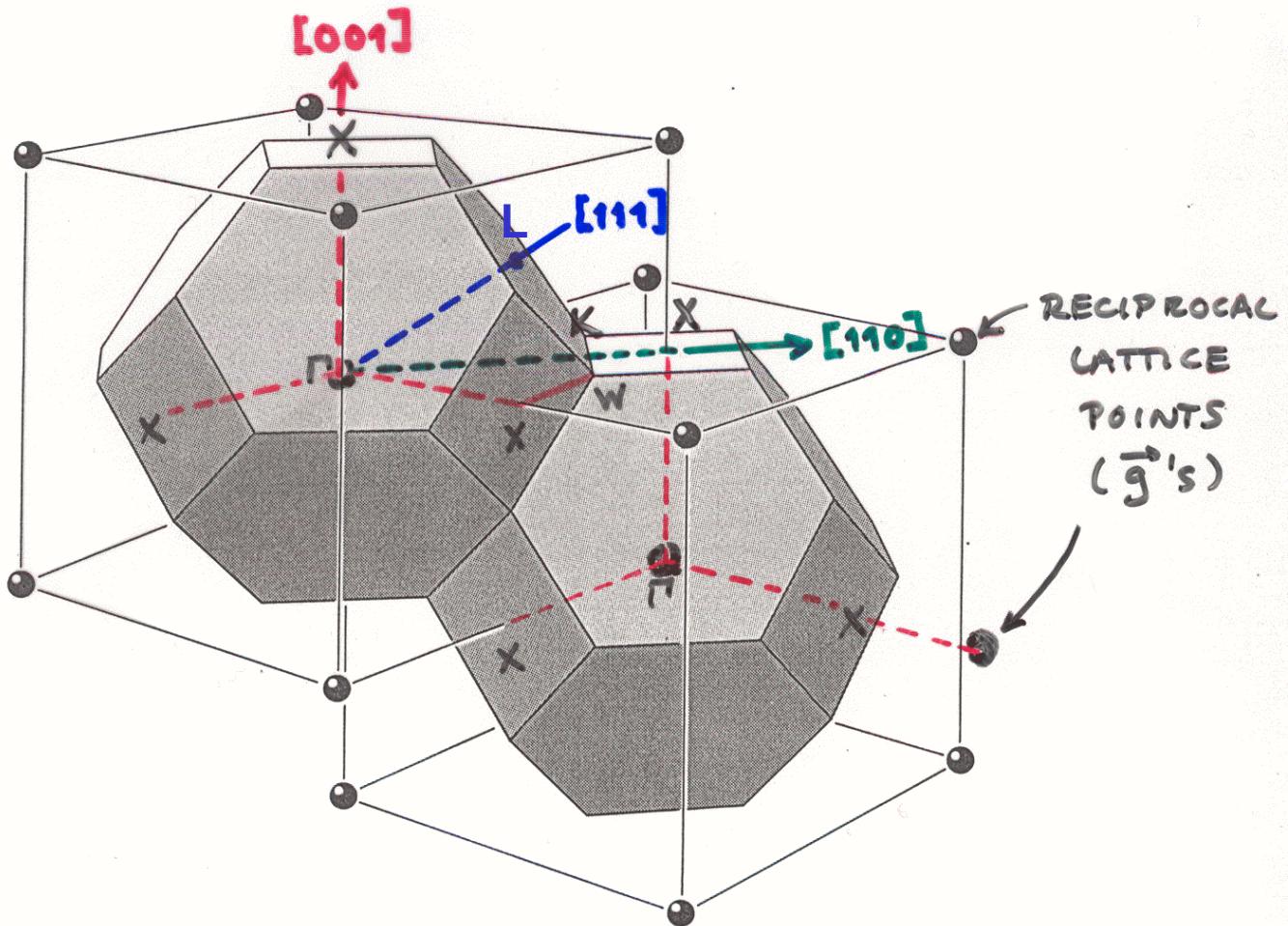


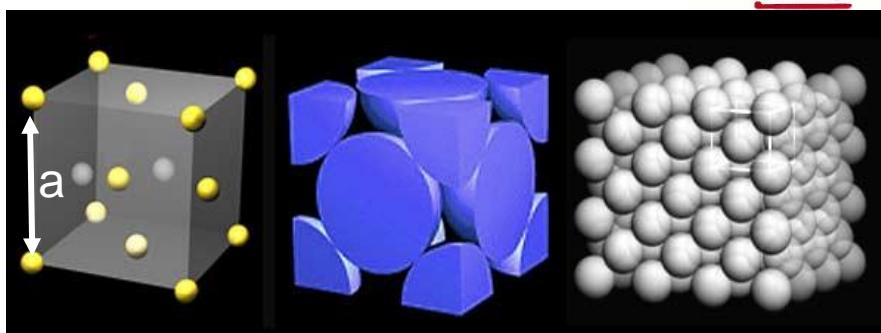
Figure 28 Brillouin zones of the face-centered cubic lattice. The cells are in reciprocal space, and the reciprocal lattice is body-centered, as drawn.

— STACKING OF fcc BRILLOUIN ZONES —

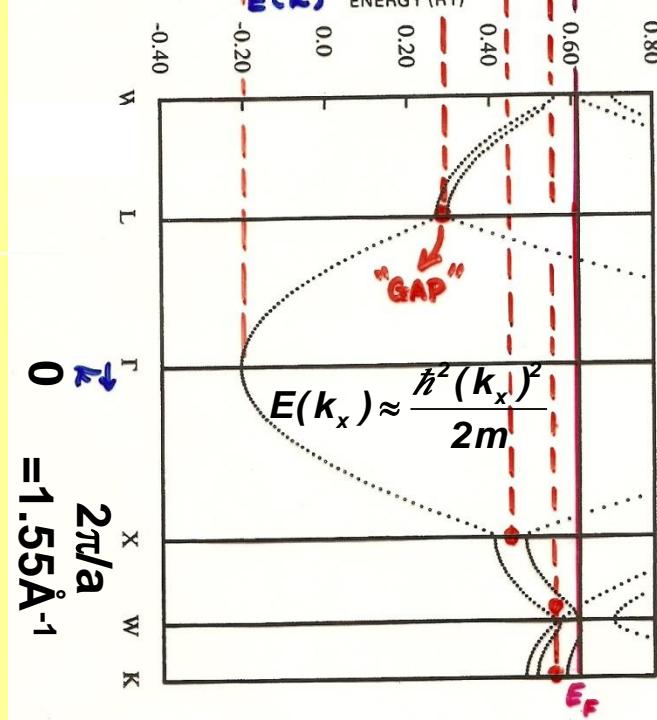
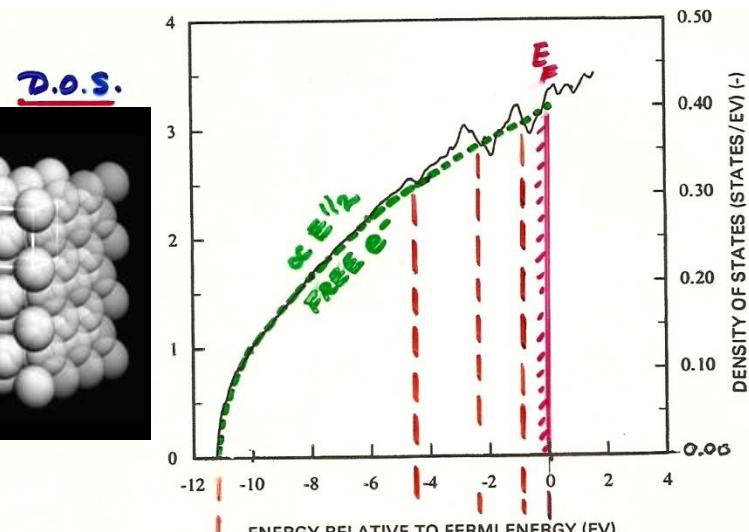
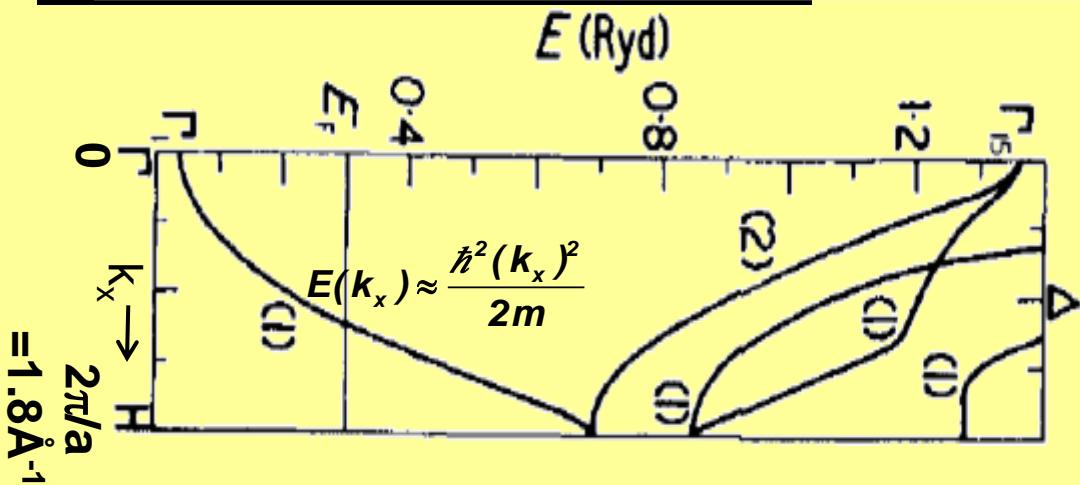
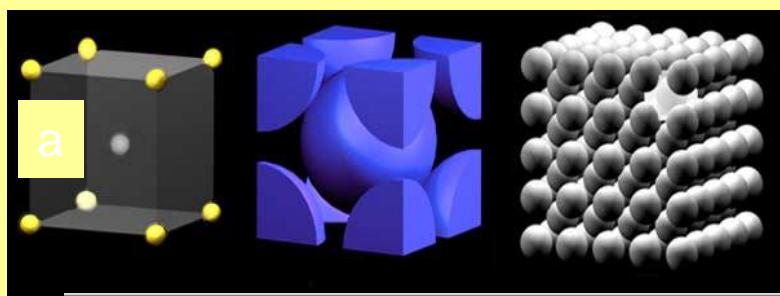
Electronic bands and density of states for “free-electron” metals -

Rydberg = 13.605 eV

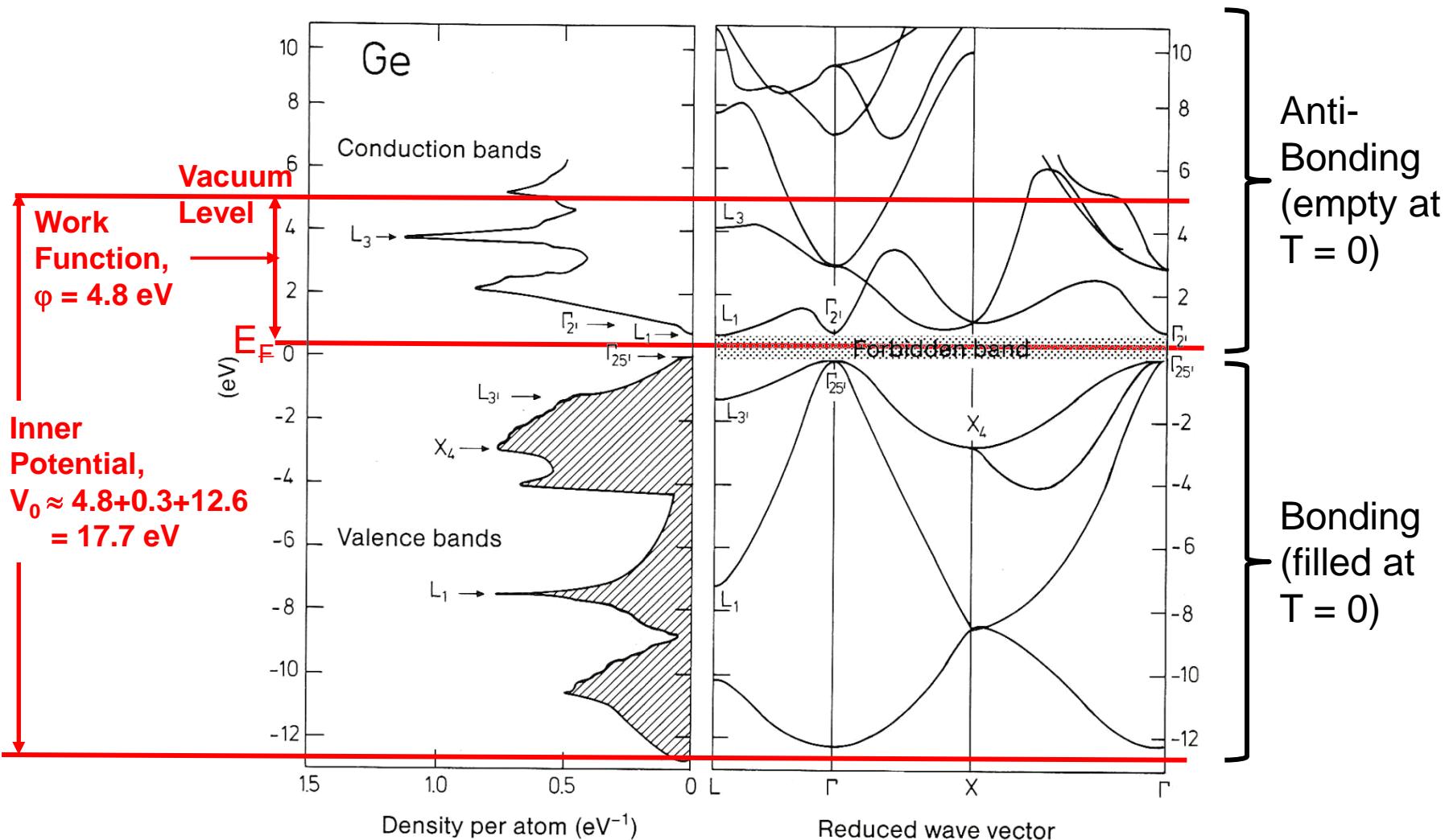
Aluminum—fcc,
 $a = 4.05 \text{ \AA}$
 $1s^2 2s^2 2p^6 3s^2 3p^1$



Lithium—bcc, $a = 3.49 \text{ \AA}$
 $1s^2 2s^1$



Electronic bands and density of states for a semiconductor-Germanium— $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} \color{red}{4s^2 4p^2}$



Vacuum level

The electronic structure of a transition metal —fcc Cu

$$\phi_{\text{Cu}} = 4.4 \text{ eV} = \text{work function}$$

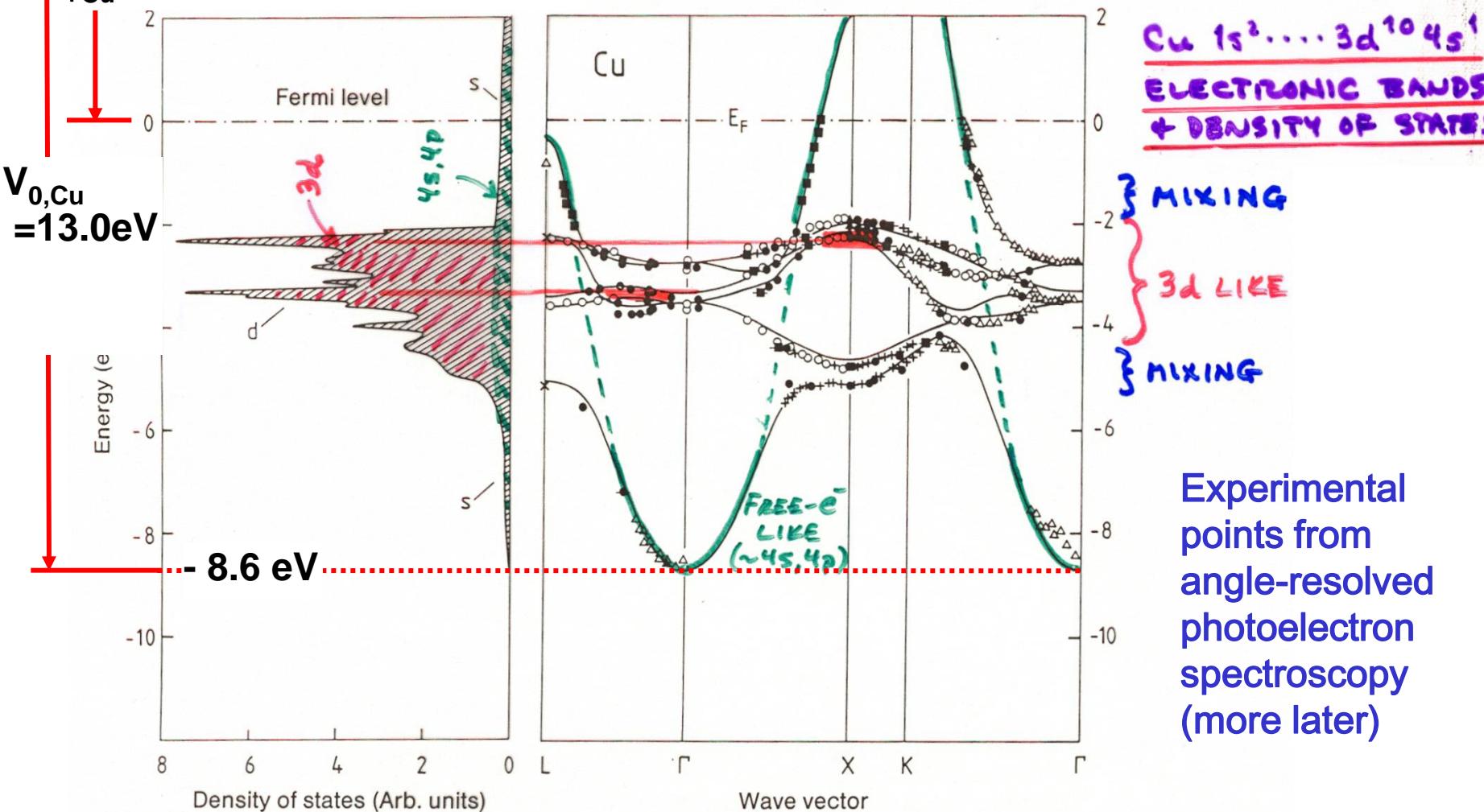


Fig. 7.12. Bandstructure $E(k)$ for copper along directions of high crystal symmetry (right). The experimental data were measured by various authors and were presented collectively by Courths and Hüfner [7.4]. The full lines showing the calculated energy bands and the density of states (left) are from [7.5]. The experimental data agree very well, not only among themselves, but also with the calculation

Atomic orbital makeup

$$\phi_j^{MO}(\vec{r}) = \sum_{\text{Atoms A}} c_{Ai,j} \phi_{Ai}^{AO}(\vec{r})$$

Atoms A
Orbitals i



5σ

Molecular orbital approach

1π_y



1π_x

4σ



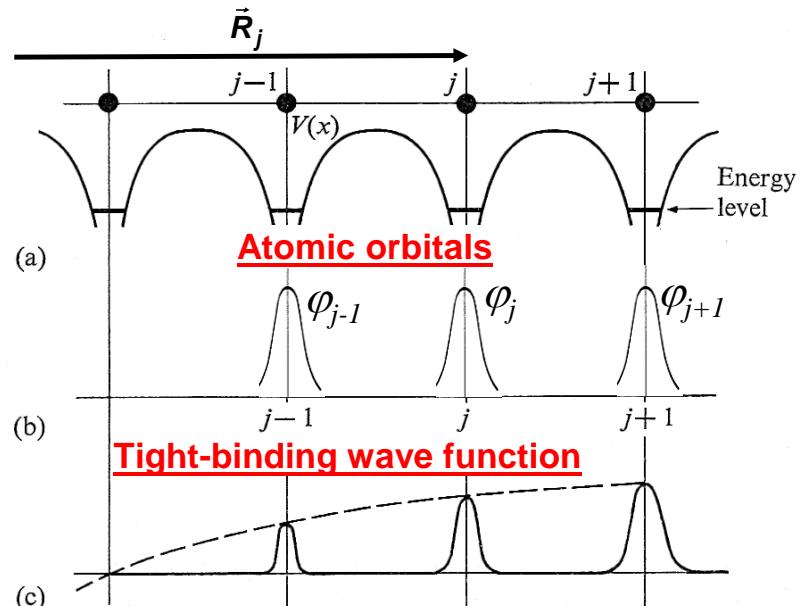
3σ

$$\phi_k^{BF}(\vec{r}) = \text{a Bloch function}$$

$$\propto \frac{1}{N^{1/2}} \sum_{j=1 \dots N \text{ unit cells at } \vec{R}_j} e^{i\vec{k} \cdot \vec{R}_j}$$

Solid state tight-binding approach

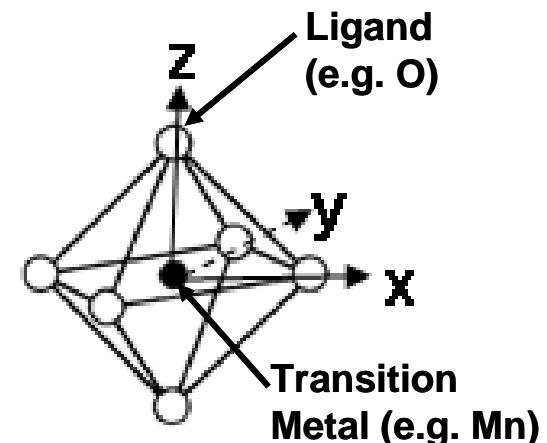
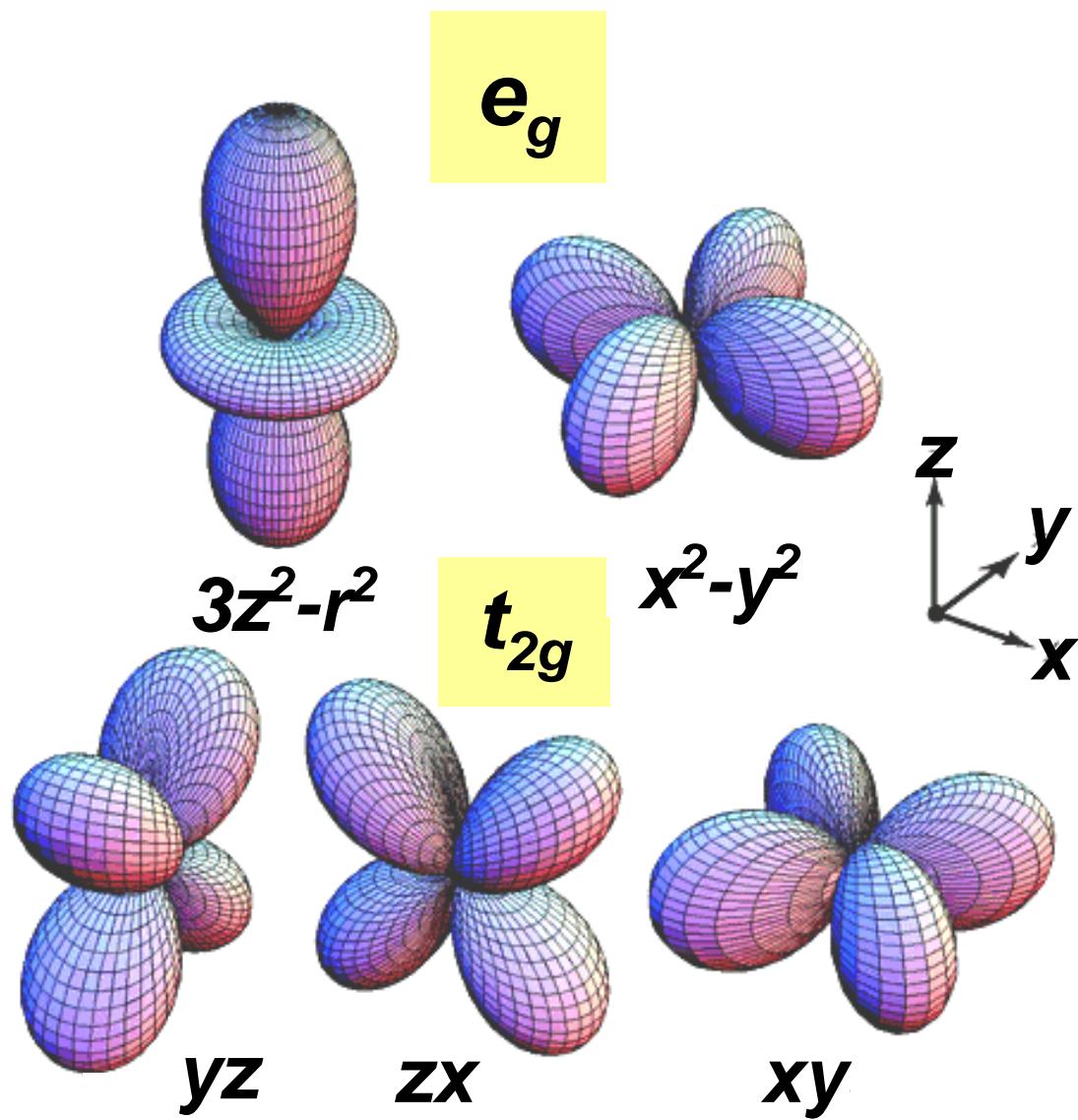
Crystal potential-1D



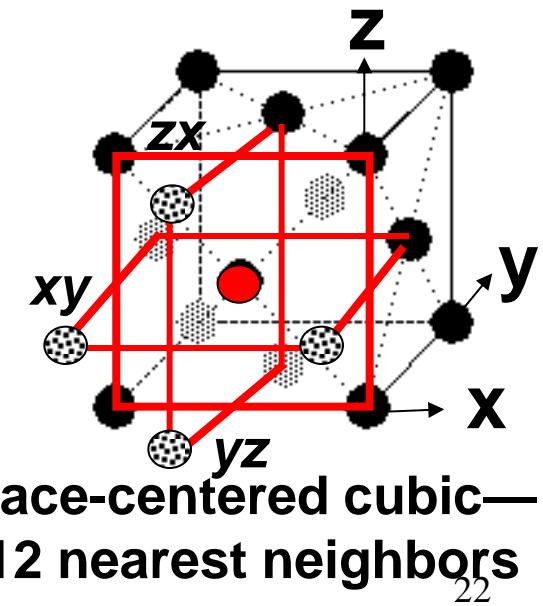
$$c_{Ai,\vec{k}} \phi_{Ai}^{AO}(\vec{r} - \vec{R}_j)_{21}$$

Ai = basis set of AOs in unit cell

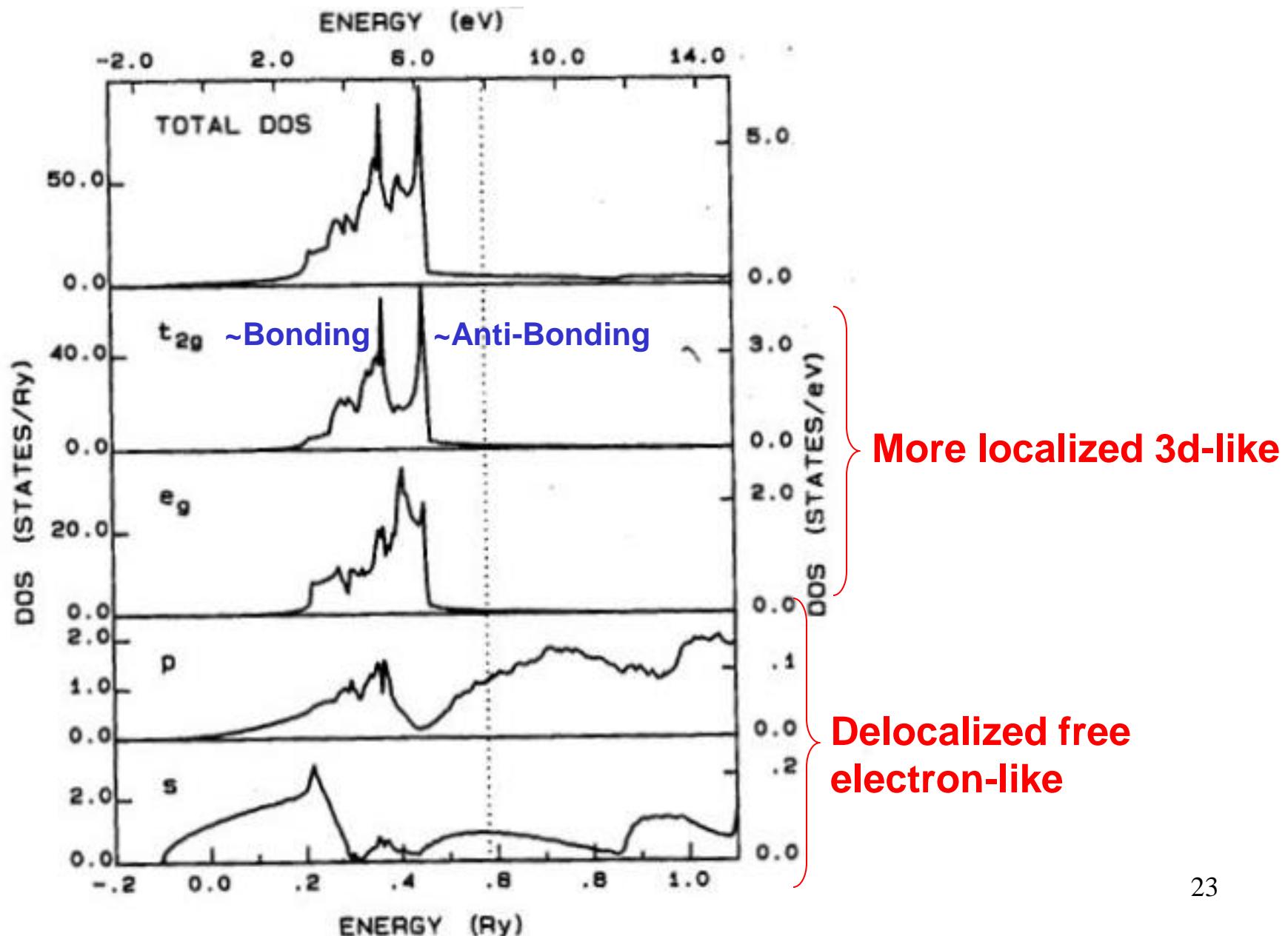
And the same thing for the d orbitals:



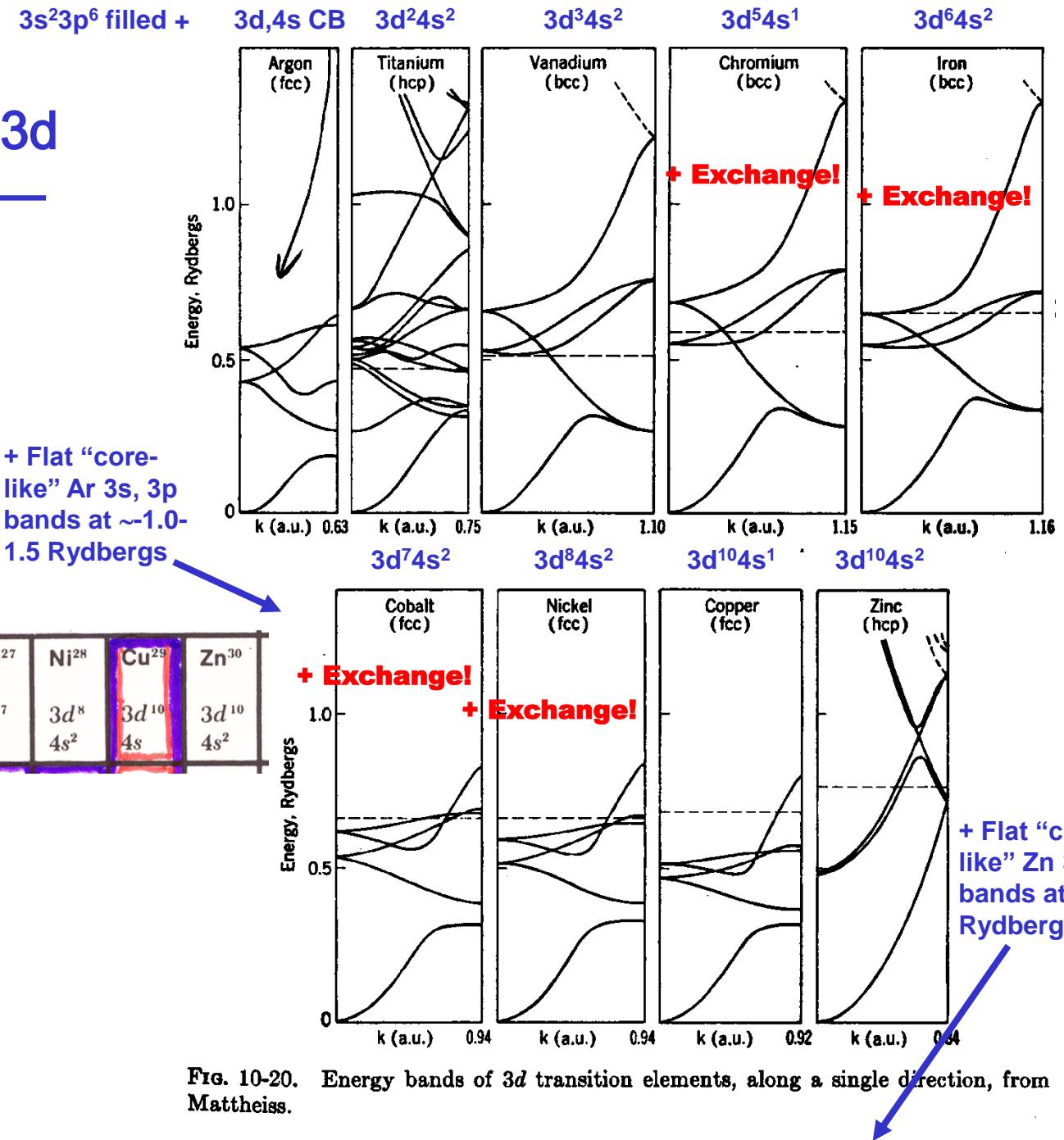
e_g and t_{2g} not equivalent in octahedral (cubic) environment



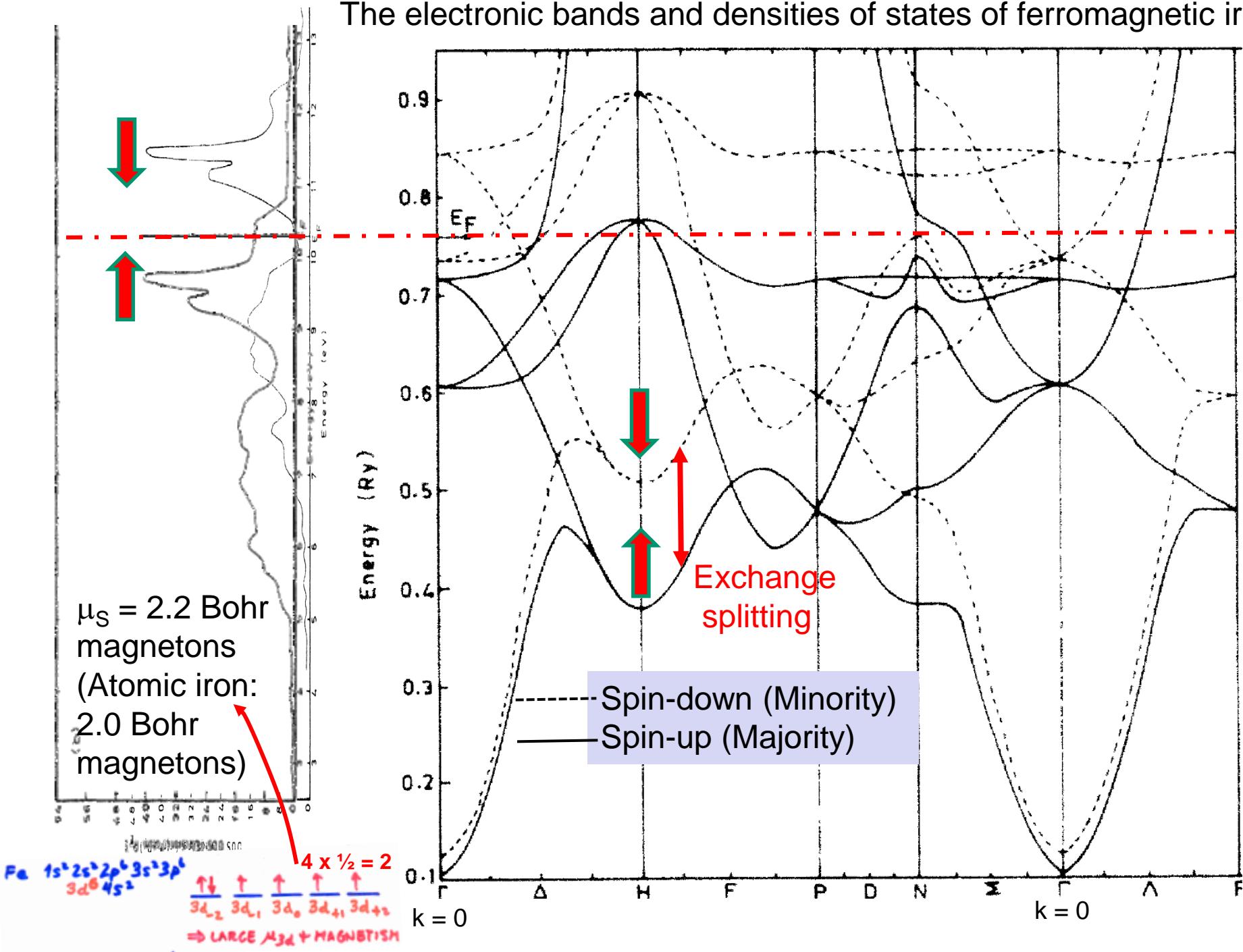
Copper densities of states-total and by orbital type:



The electronic structures of the 3d transition metals— ≈ “rigid-band model”



The electronic bands and densities of states of ferromagnetic iron



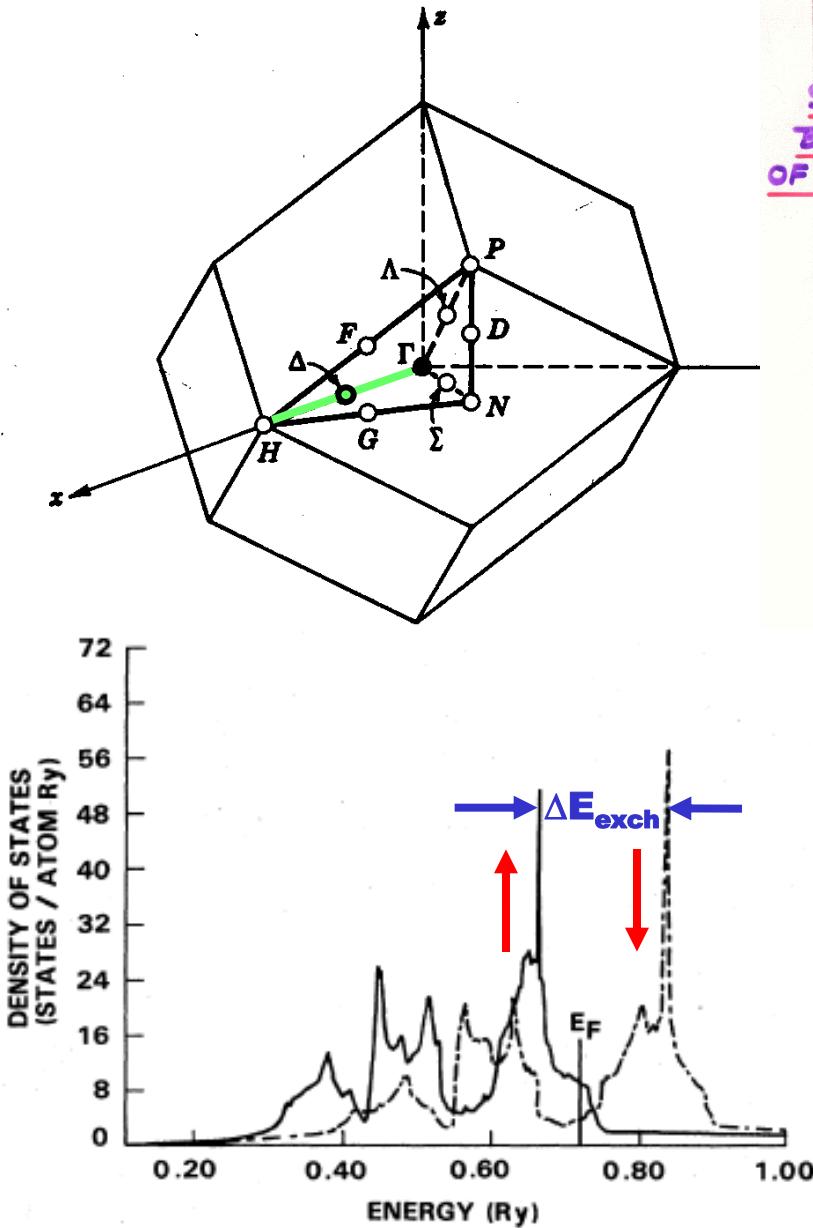
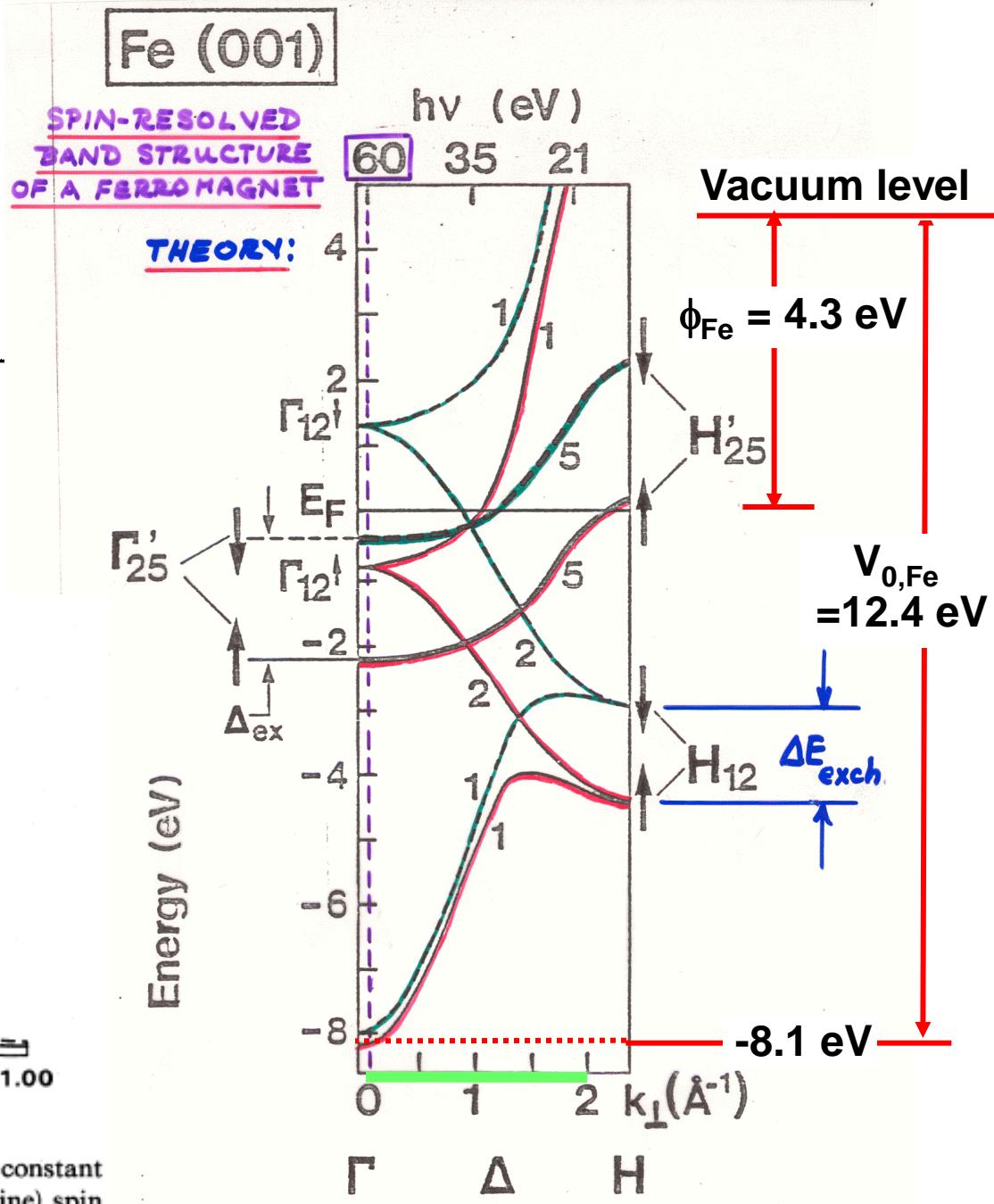
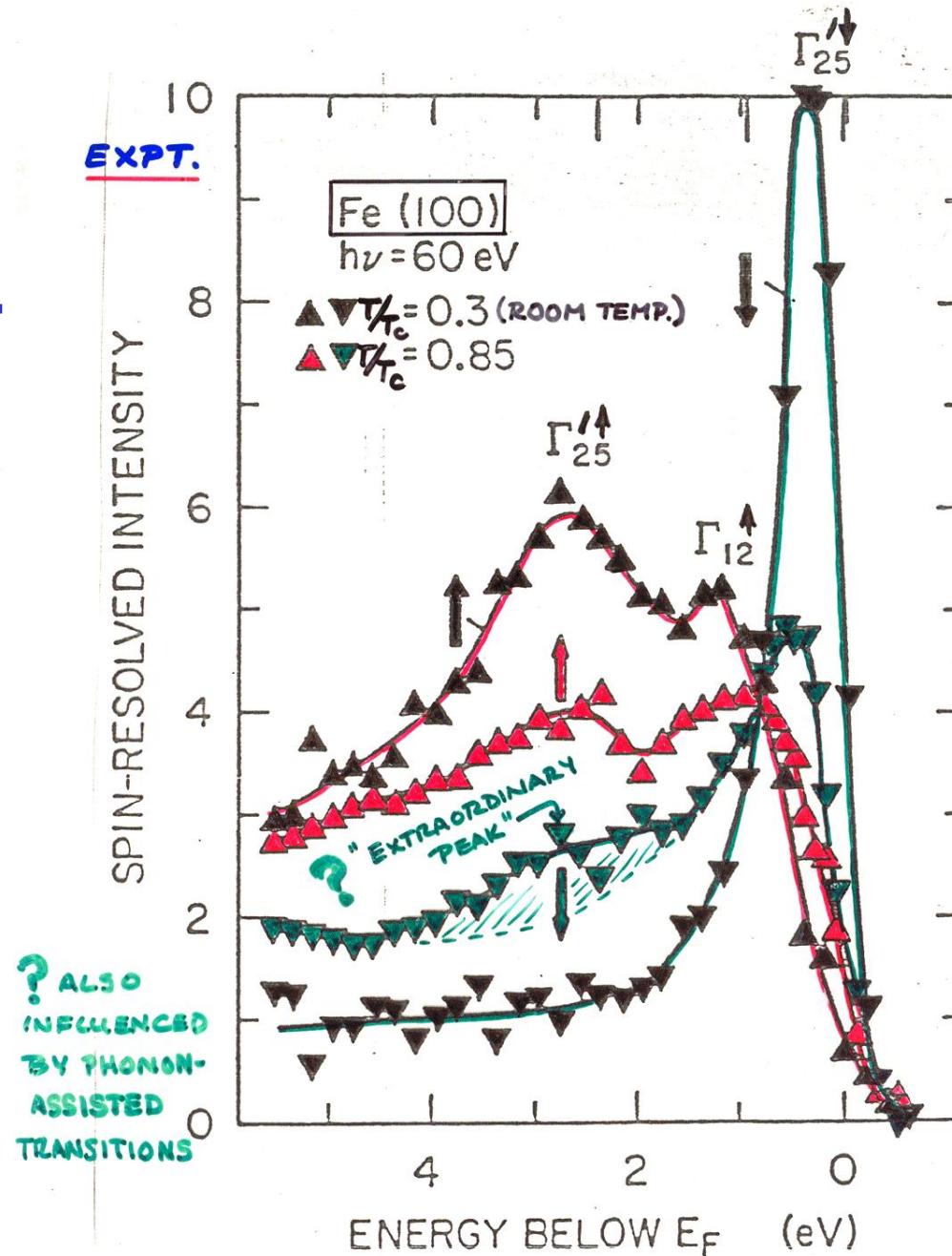


FIG. 4. Density of states at the equilibrium lattice constant of Fe for majority- (solid line) and minority- (broken line) spin states.

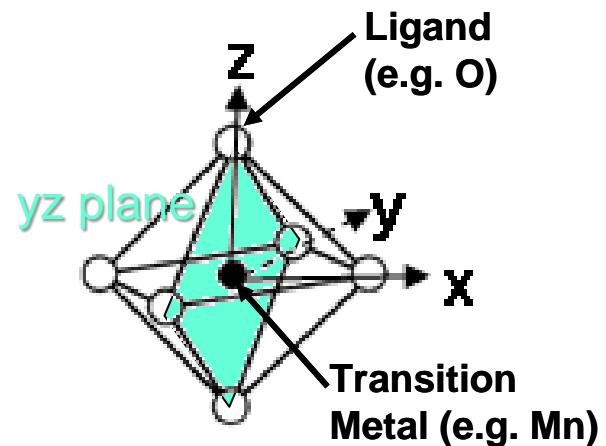
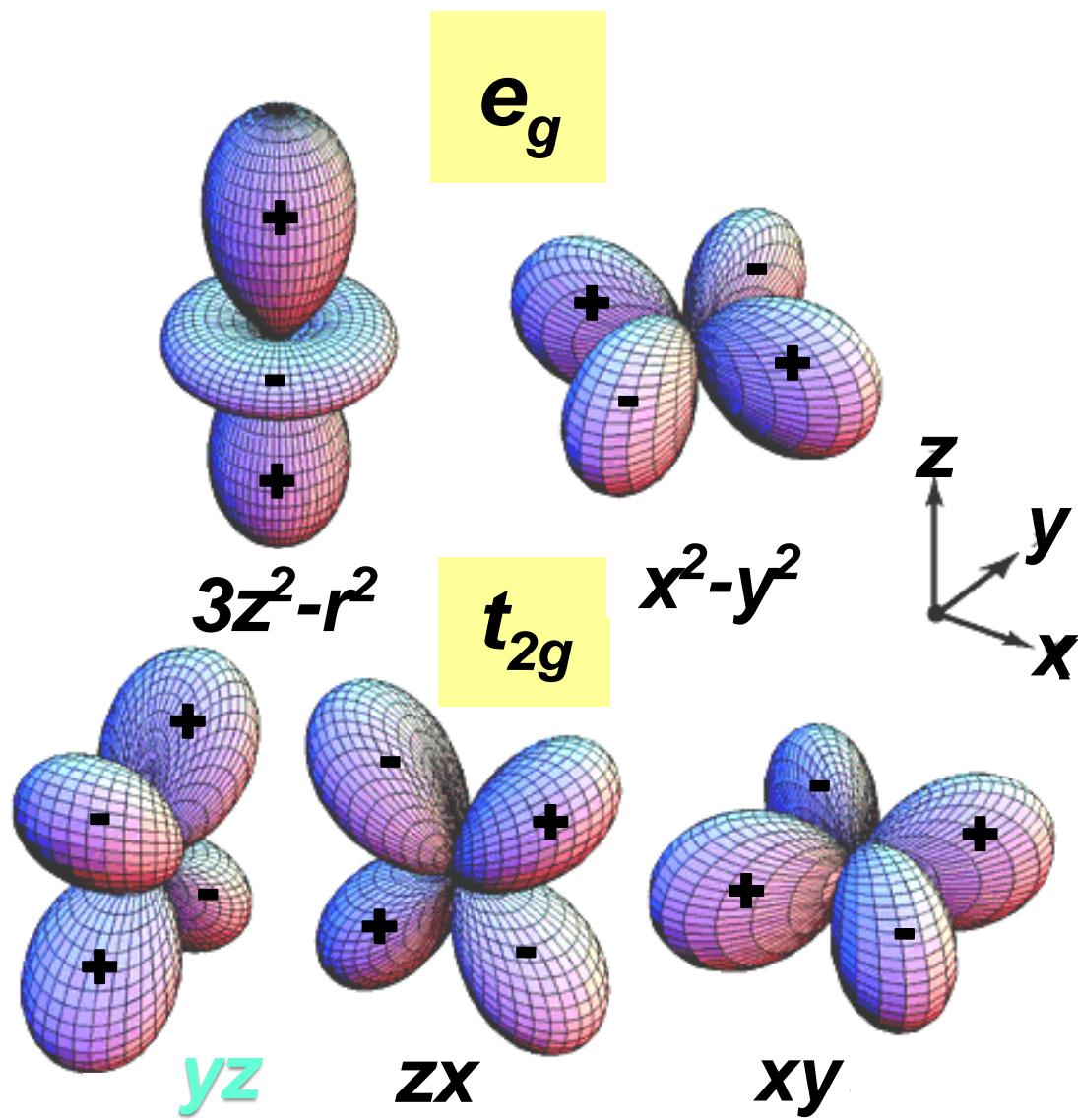
Hathaway et al., Phys. Rev. B 31, 7603 ('85)



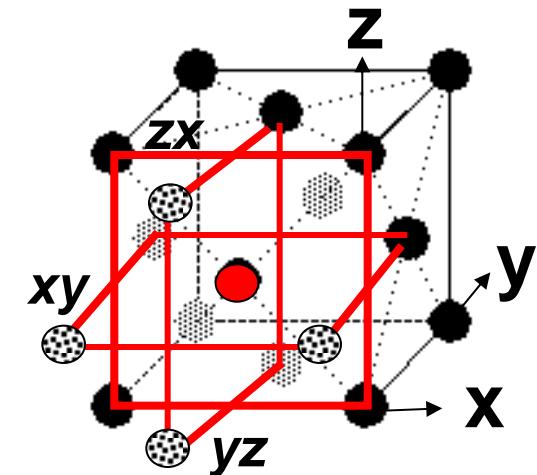
Fe: ANGLE AND SPIN-RESOLVED SPECTRA AT Γ POINT



And the same thing for the d orbitals:

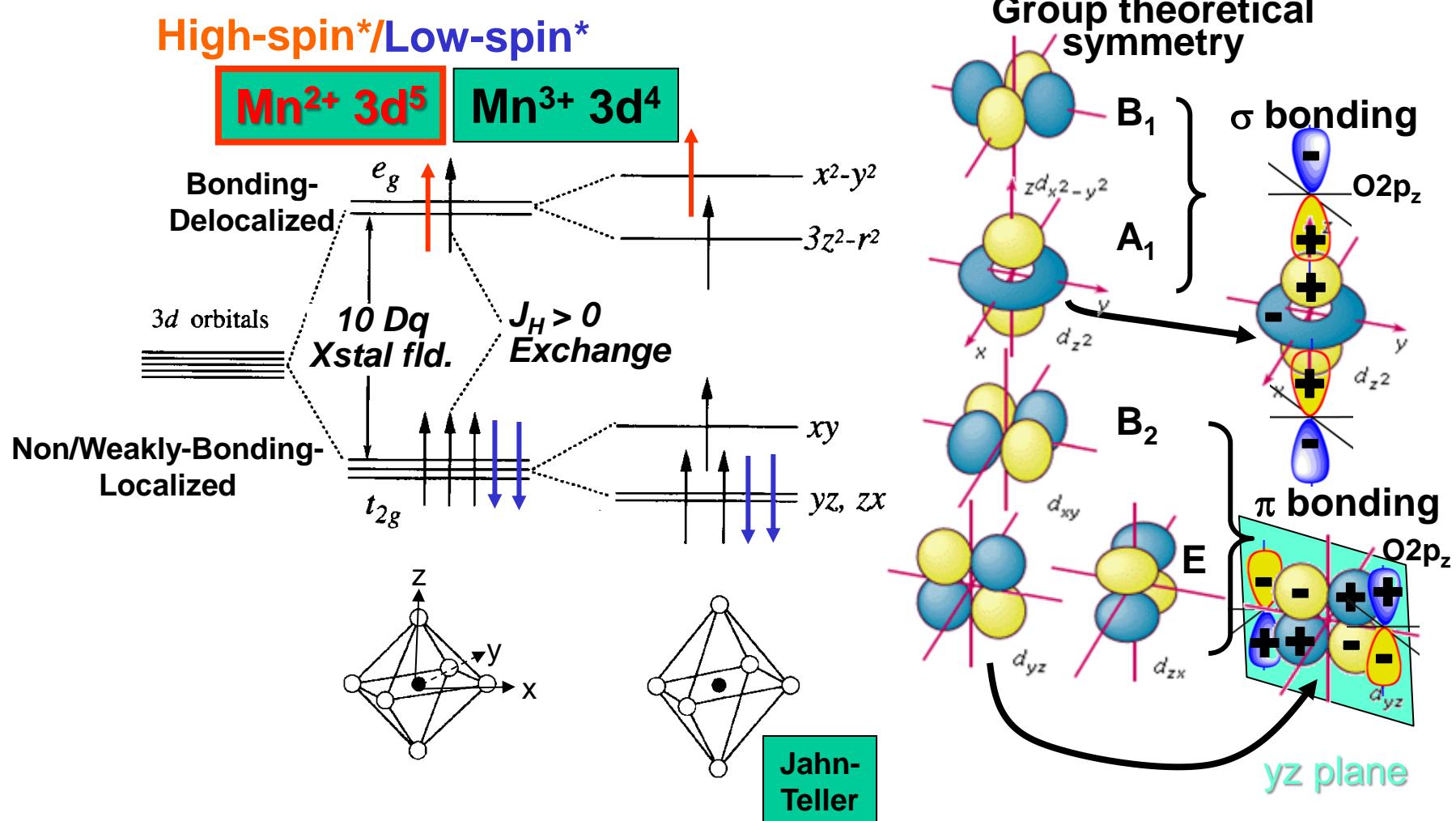


e_g and t_{2g} not equivalent in octahedral (cubic) environment



Face-centered cubic—
12 nearest neighbors

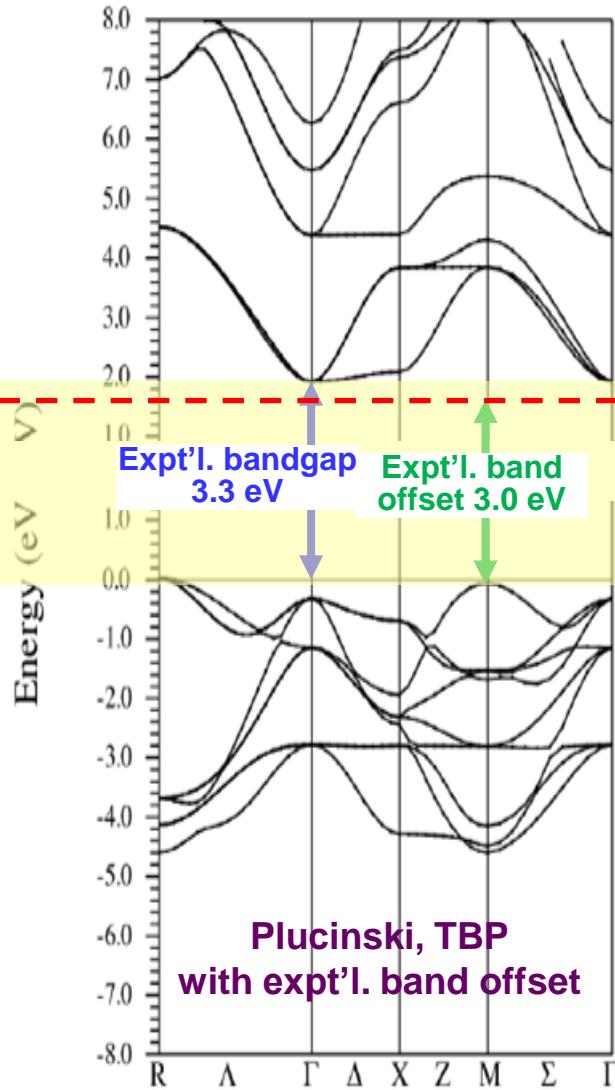
E.g.—Crystal field in Mn^{3+} & Mn^{2+} with negative octahedral ligands



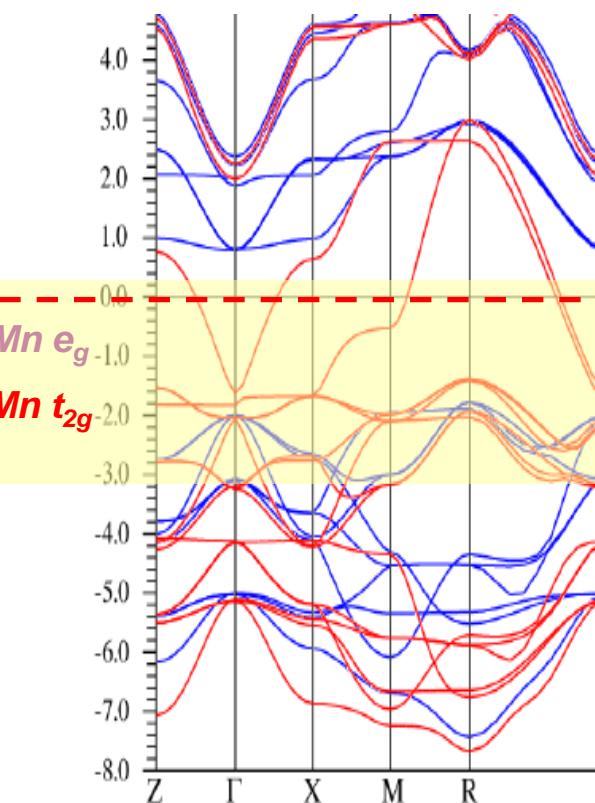
High-spin*: $10Dq \ll J_H$
 Low-spin*: $10Dq \gg J_H$

SrTiO₃ and La_{0.67}Sr_{0.33}MnO₃ band structures and DOS

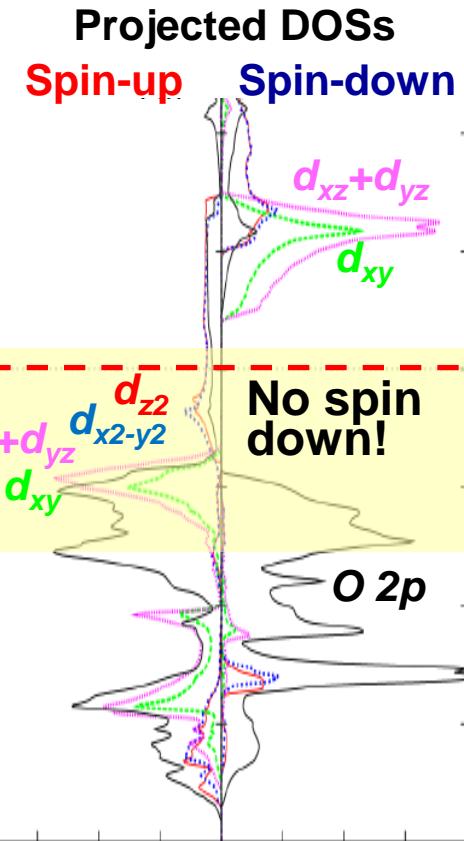
SrTiO₃-band insulator



La_{0.67}Sr_{0.33}MnO₃- Half-Metallic Ferromagnet

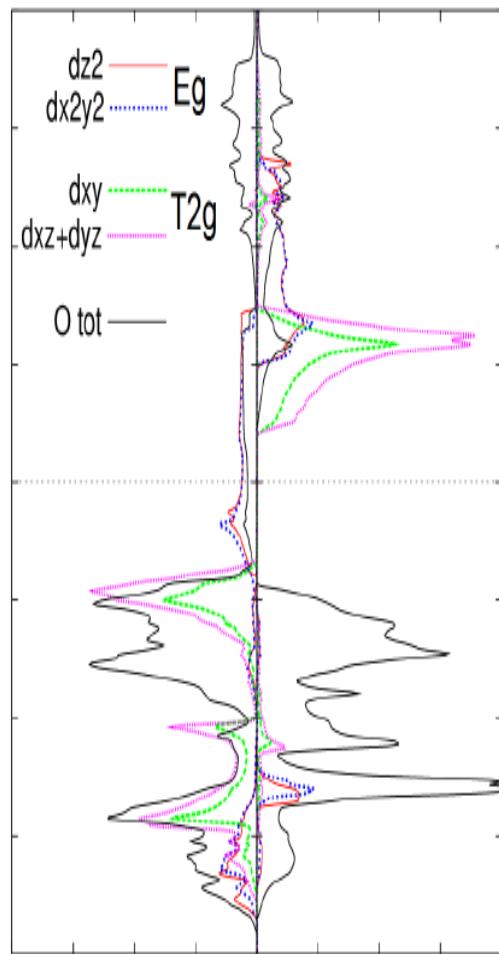


Spin-up
Spin-down
Chikamatsu et al.,
PRB 73, 195105 (2006);
Plucinski, TBP



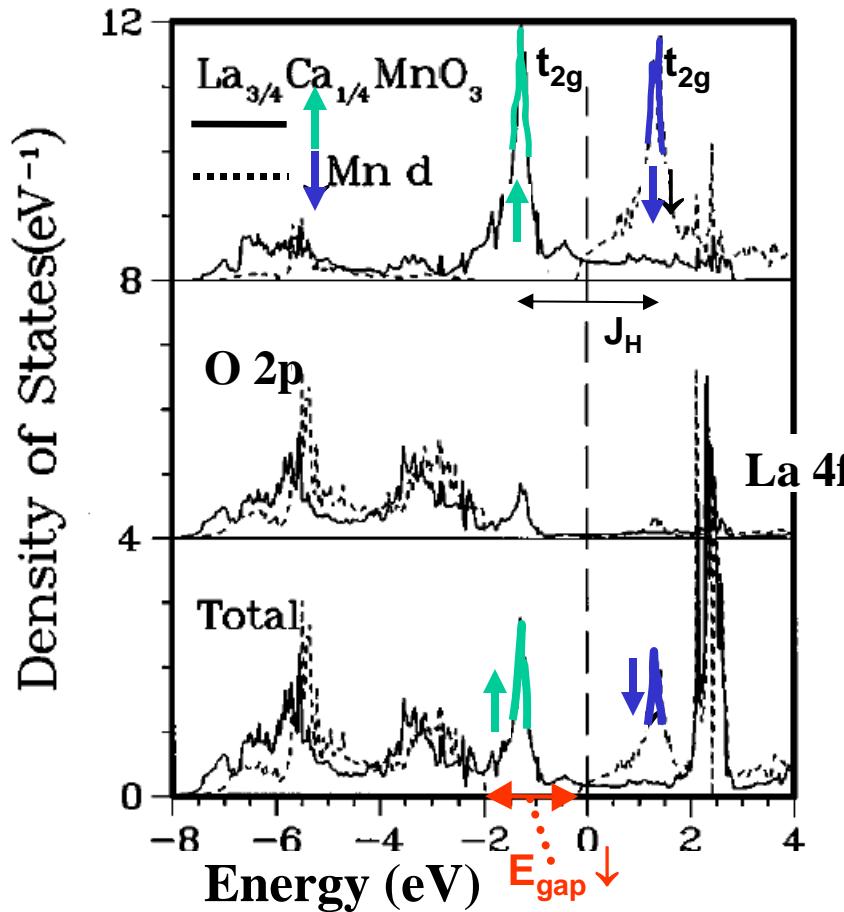
Zheng, Binggeli, J. Phys.
Cond. Matt. 21, 115602 (2009)
Plucinski, TBP

LSMO partial DOS



Half-Metallic Ferromagnetism

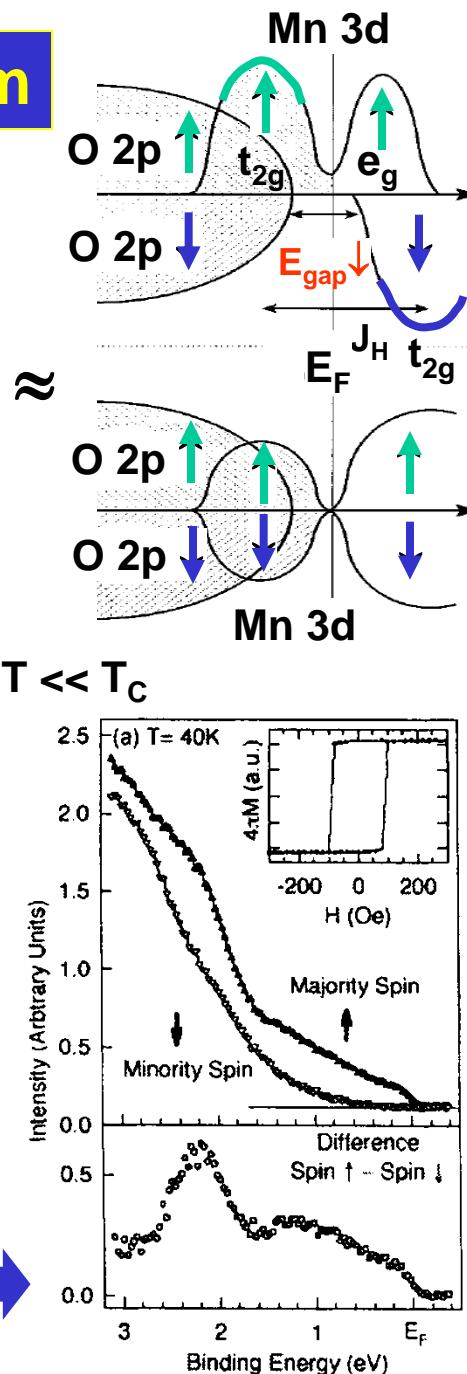
LDA theory- FM $\text{La}_{0.75}\text{Ca}_{0.25}\text{MnO}_3$



Pickett and Singh, PRB 53, 1146 (1996)

Experiment- spin-resolved PS
 $\text{La}_{0.70}\text{Sr}_{0.30}\text{MnO}_3$ as thin film

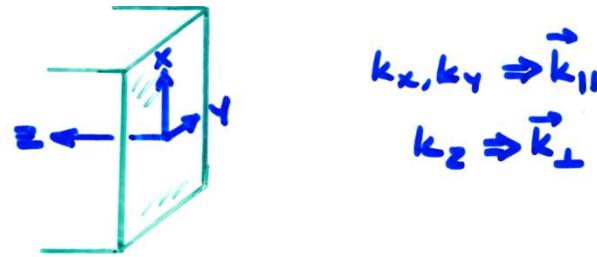
Park et al., Nature, PRB 392, 794 (1998)



FM : $T \ll T_c$

PM : $T > T_c$

SURFACE ELECTRONIC STATES

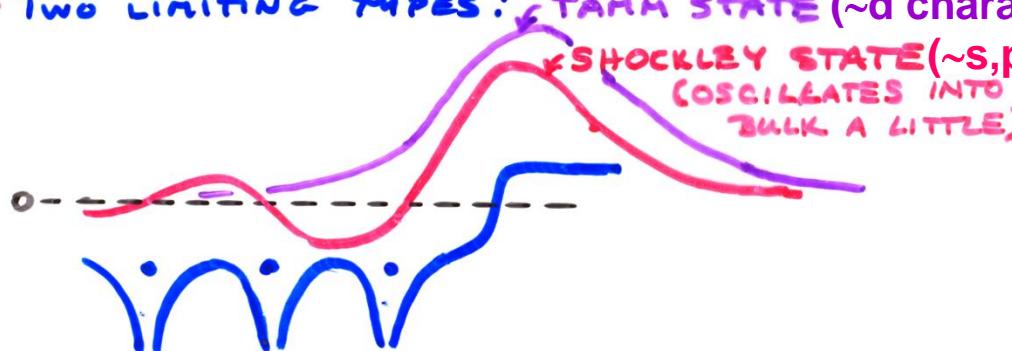


- STRONGLY LOCALIZED NEAR SURFACE
- BLOCH FUNCTION IN $x+y$, BUT DECAYING IN z :

$$\varphi_{\vec{k}_{\parallel}}(\vec{r}) \approx u_{\vec{k}_{\parallel}}(\vec{r}) e^{i \vec{k}_{\parallel} \cdot \vec{r}} e^{-k_z z}$$

↑
DECAY CONSTANT

- Two limiting types:
 - TAMM STATE (~d character, localized)
 - SHOCKLEY STATE (~s,p character, delocalized)
(OSCILLATES INTO BULK A LITTLE)



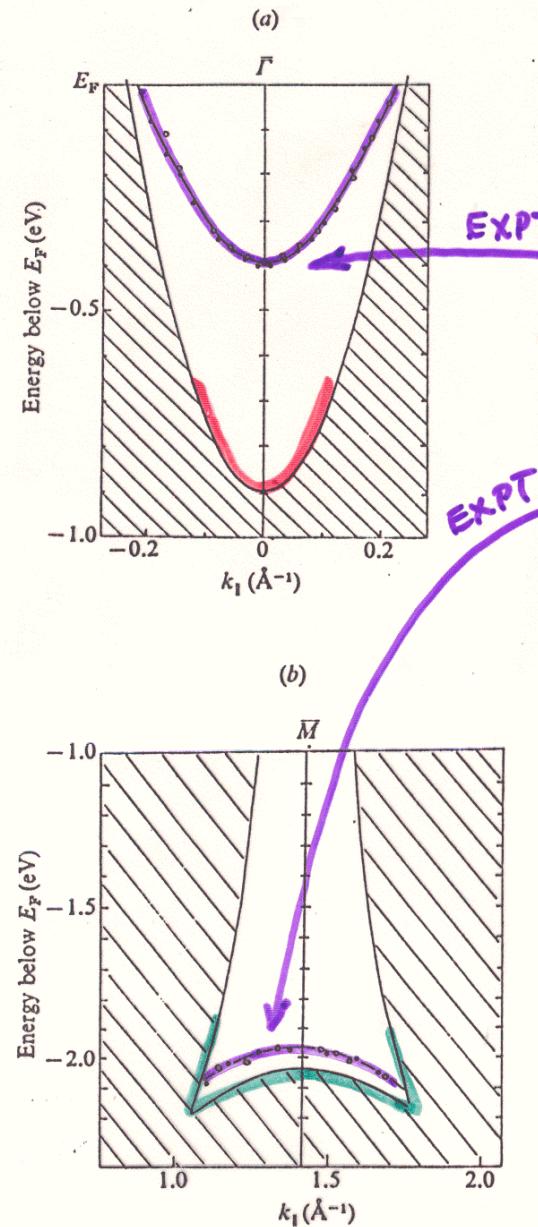
- ONLY EXIST WHEN NO BULK STATE EXISTS AT SAME $E_{\parallel} = k_x \hat{i} + k_y \hat{j}$; OTHERWISE MIXING OCCURS + NOT SURFACE-LOCALIZED

Surface states on Cu(111)

**Shockley
surface
State:
s,p makeup**

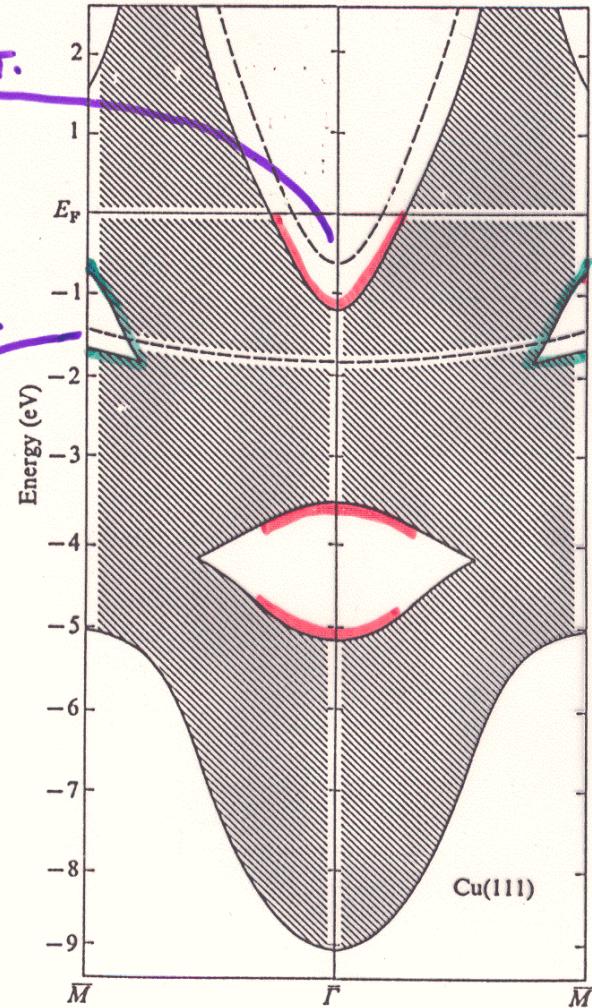
**Tamm
surface
state:
3d makeup**

Fig. 4.21. Experimental dispersion of Cu(111) surface states plotted with a projection of the bulk bands: (a) Shockley state near the zone center (Kevan, 1983); (b) Tamm state near the zone boundary (Heimann, Hermanson, Miosga and Neddermeyer, 1979). Compare with Fig. 4.17.



THEORY

Fig. 4.17. Surface states (dashed curves) and bulk projected bands for Cu(111) surface according to a six-layer surface band structure calculation (Euceda, Bylander & Kleinman, 1983).



CONSERVATION LAWS IN VALENCE-BAND PHOTOELECTRON SPECTROSCOPY!

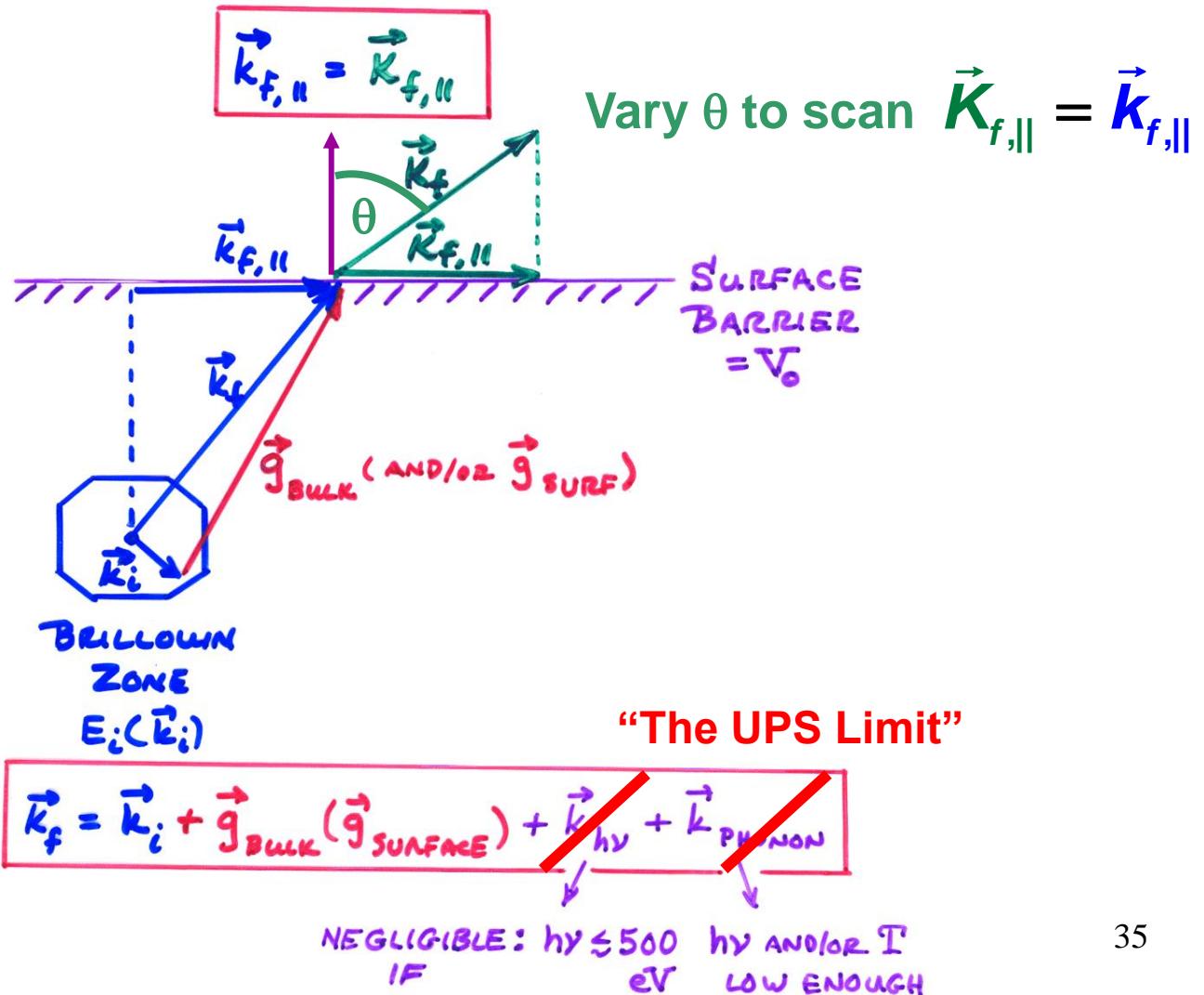
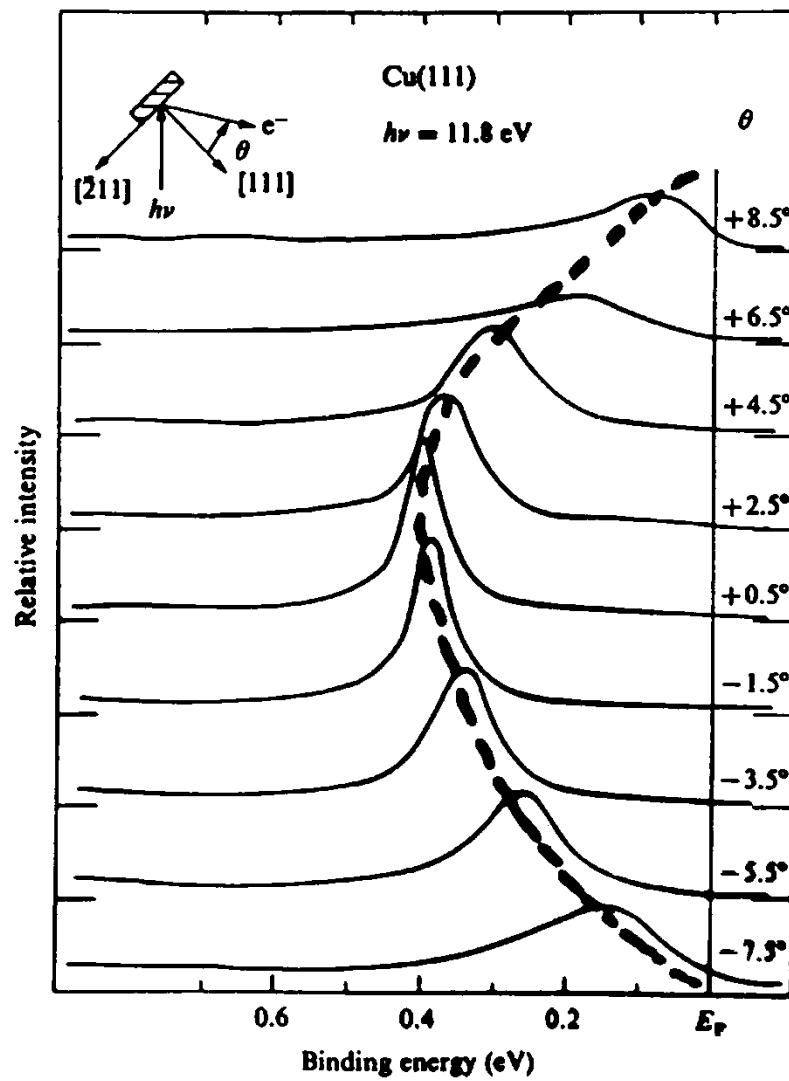
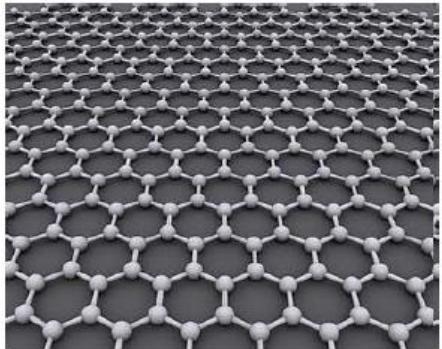


Fig. 4.20. Photoemission energy distribution curves from Cu(111) at different collection angles. Equation (4.32) has been used to express the electron kinetic energy in terms of the binding energy of the electron state (Kevan, 1983).

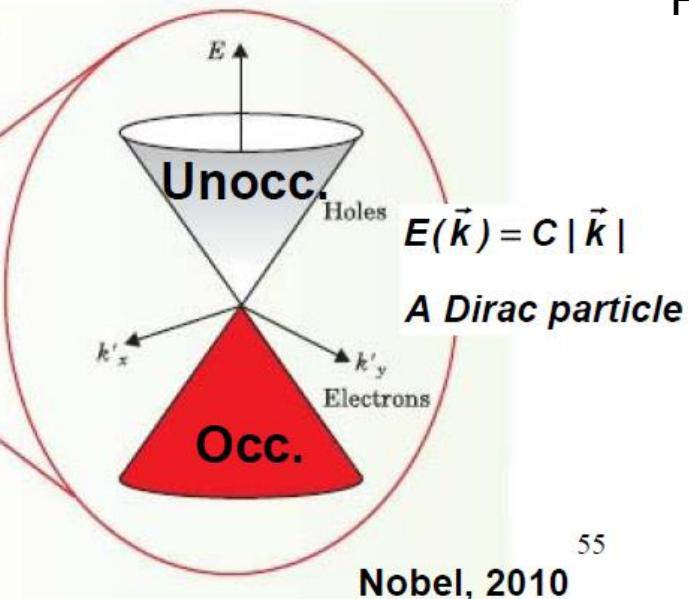
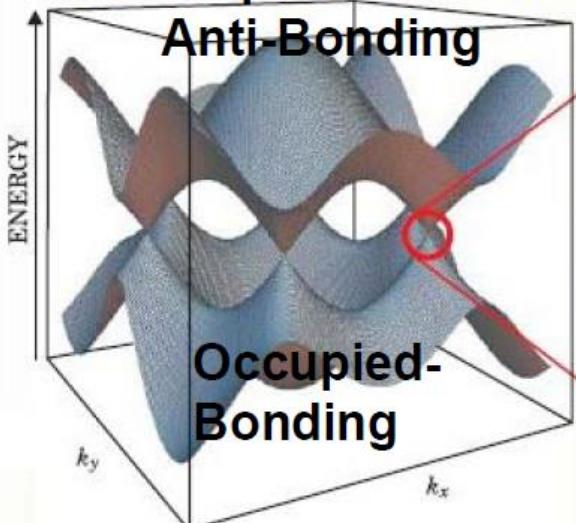


Graphene- A very special 2D case

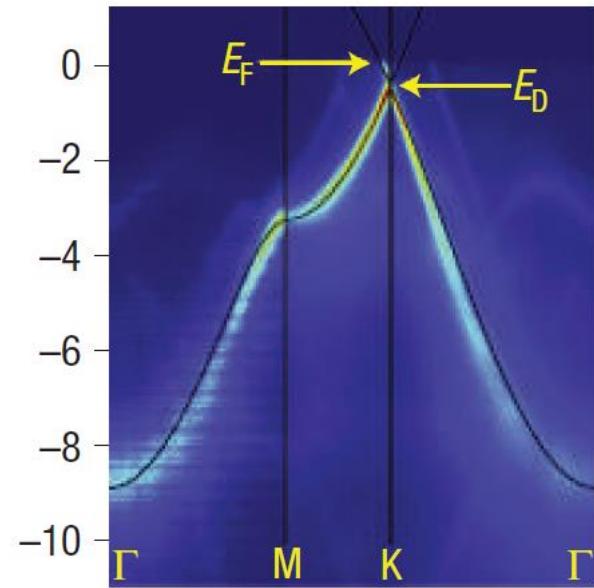


The Nobel Prize in Physics 2010
Andre Geim, Konstantin Novoselov
... "for groundbreaking experiments
regarding the two-dimensional
material graphene"

Unoccupied at T = 0K-

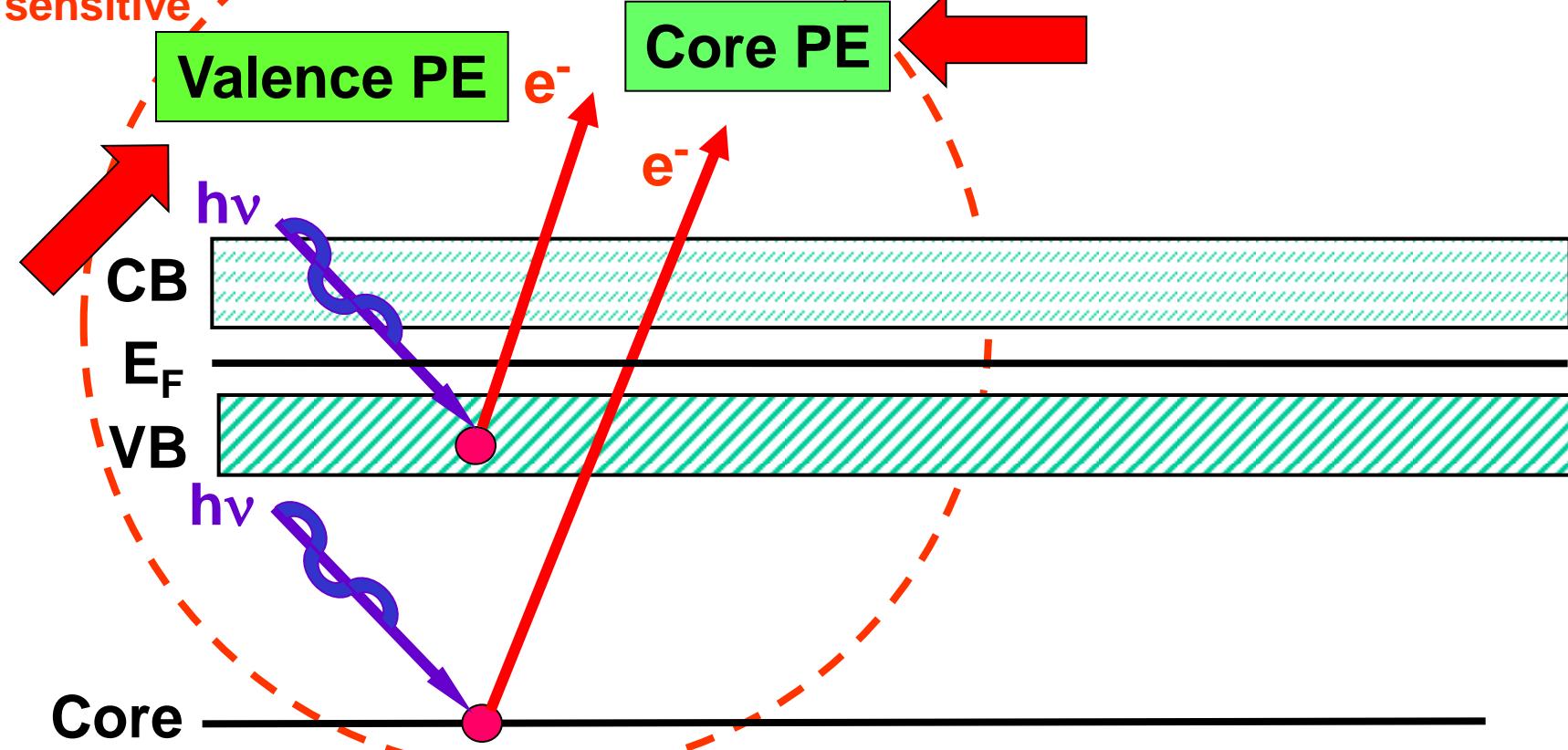


Photoelectron spectroscopy



The Soft X-Ray Spectroscopies

Electron-out:
surface
sensitive



PE = photoemission = photoelectron spectroscopy

XAS = x-ray absorption spectroscopy

AES = Auger electron spectroscopy

XES = x-ray emission spectroscopy

REXS/RIXS = resonant elastic/inelastic x-ray scattering

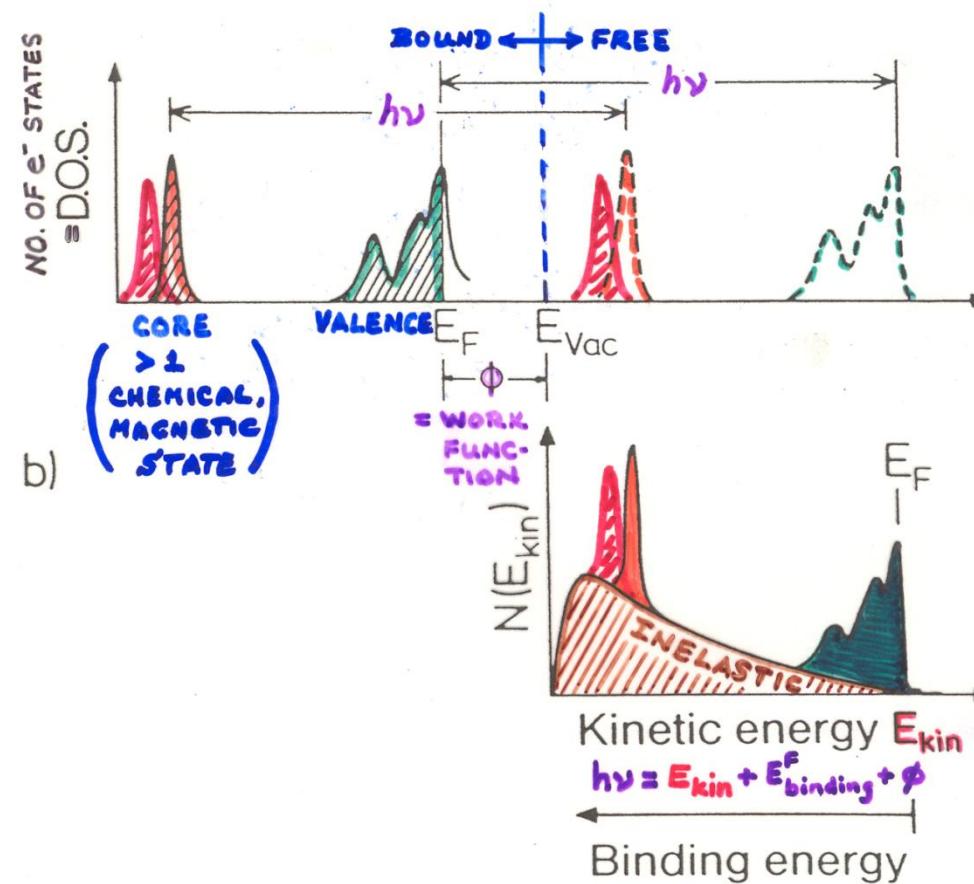
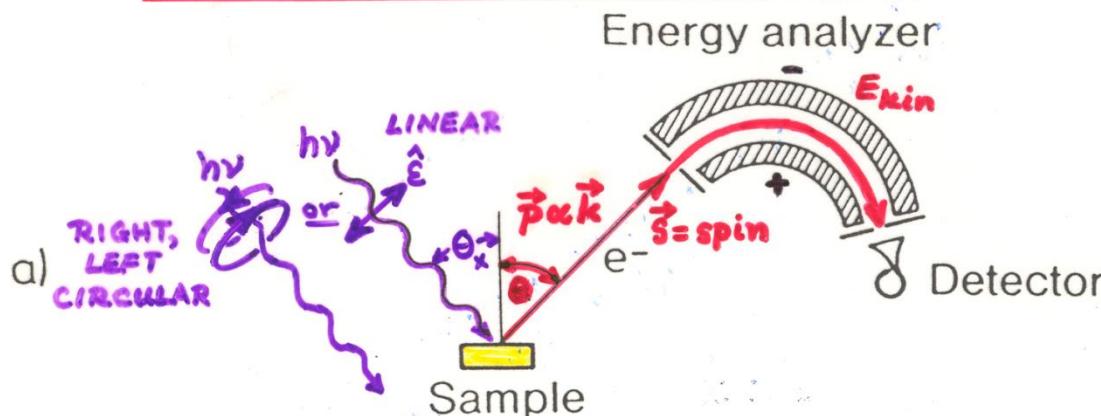
MATRIX ELEMENTS IN THE SOFT X-RAY SPECTROSCOPIES: DIPOLE LIMIT

- Photoelectron spectroscopy/photoemission:

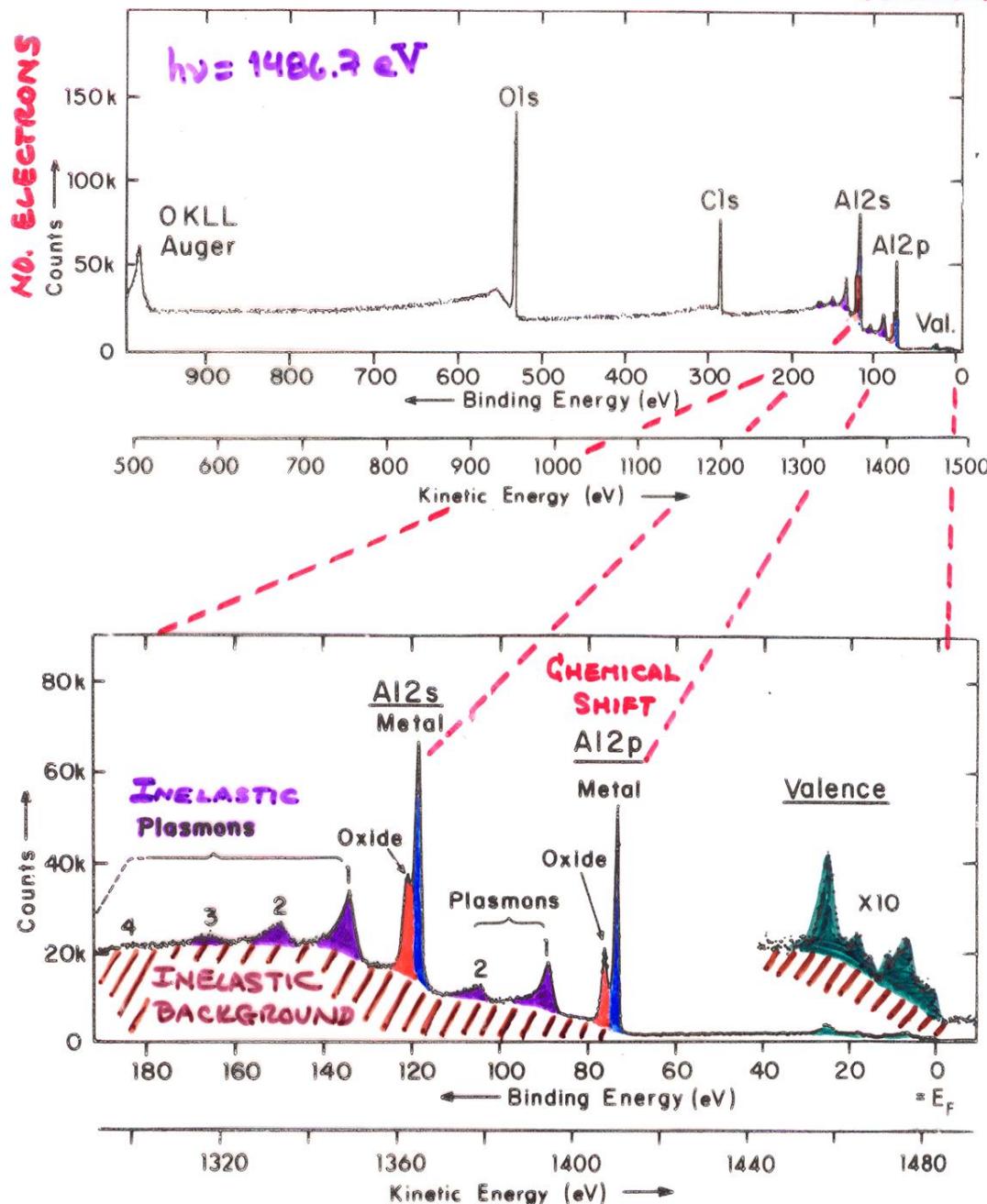
$$I \propto |\hat{\mathbf{e}} \cdot \langle \varphi_f(1) | \vec{r} | \varphi_i(1) \rangle|^2$$

The diagram illustrates the photoemission process. A purple wavy arrow labeled $h\nu$ represents an incoming photon. It strikes a red dot labeled $\varphi_i(\text{bound})$, which represents a bound electron. From this point, a red arrow labeled $\varphi_f(\text{free})$ extends upwards, representing the free electron after emission. A dashed line labeled "Vacuum" separates the interaction region from the free space above.

PHOTOELECTRON SPECTROSCOPY



TYPICAL PHOTOELECTRON SPECTRA: OXIDIZED ALUMINUM



"Basic Concepts of XPS"
Figure 1

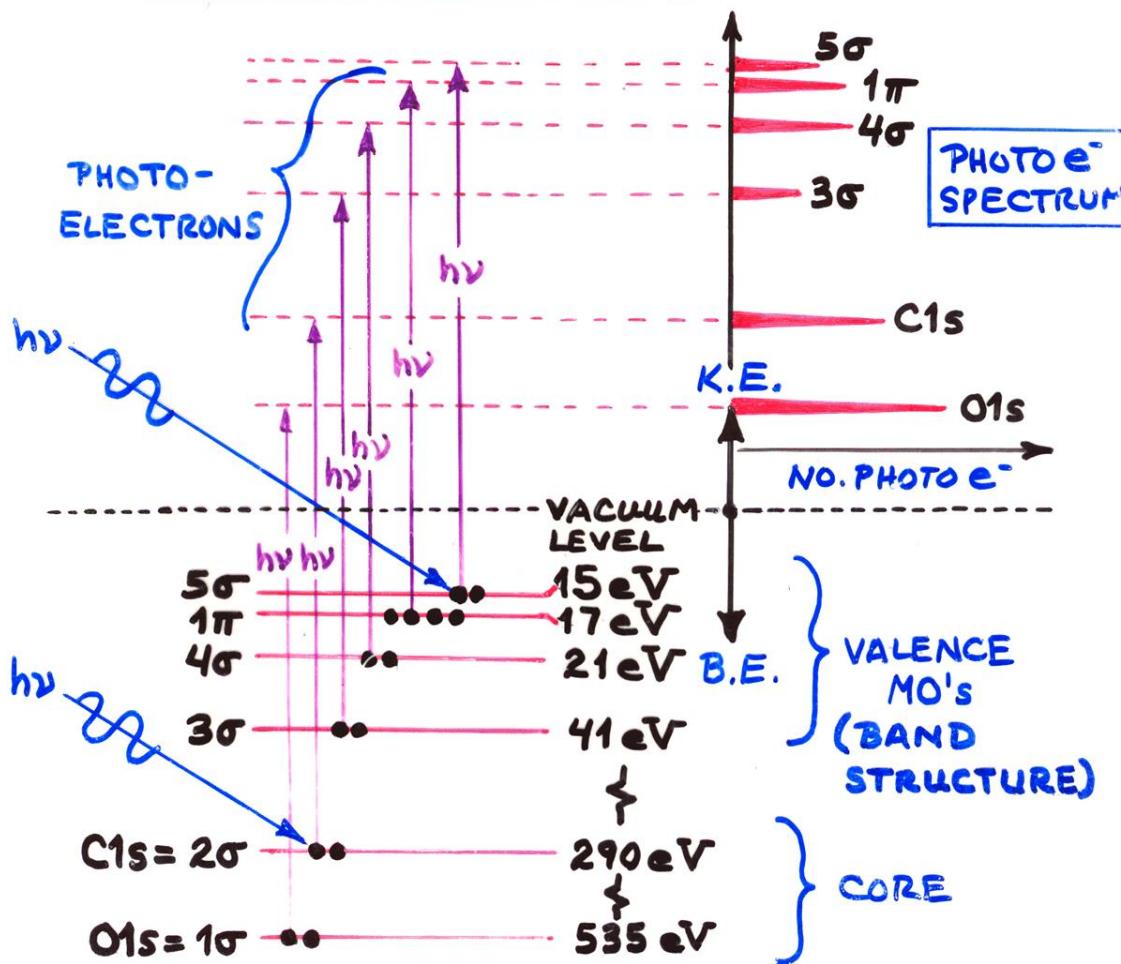
PHOTOELECTRON SPECTROSCOPY

THE PHOTOELECTRIC EFFECT (EINSTEIN, 1905):

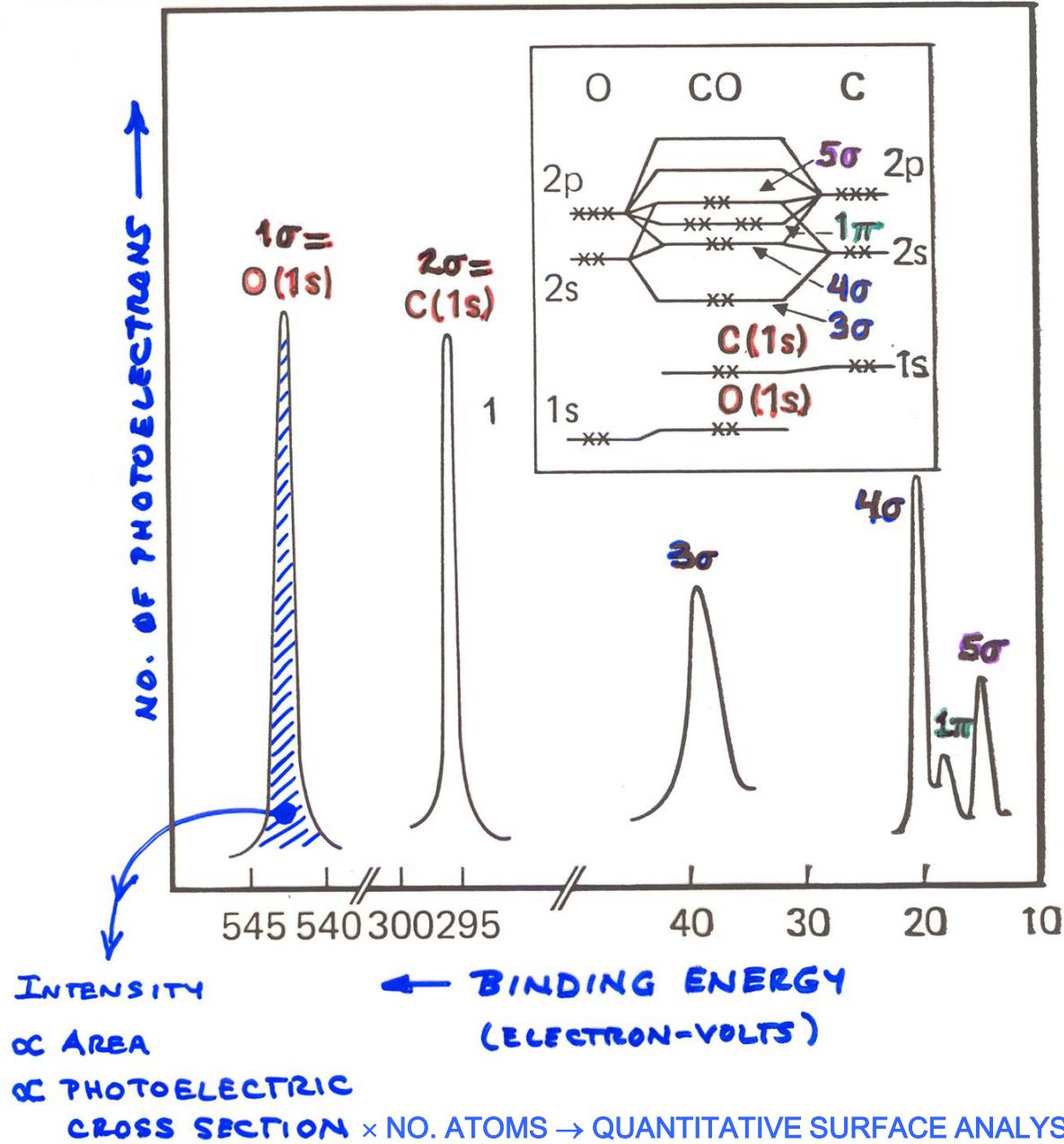
$$(\text{PHOTON ENERGY}) = (\text{e}^- \text{ BINDING ENERGY IN SYSTEM}) + (\text{PHOTOELECTRON KINETIC ENERGY EMITTED})$$

(ABSORBED) = B.E. + K.E.

EXAMPLE - CO MOLECULE:

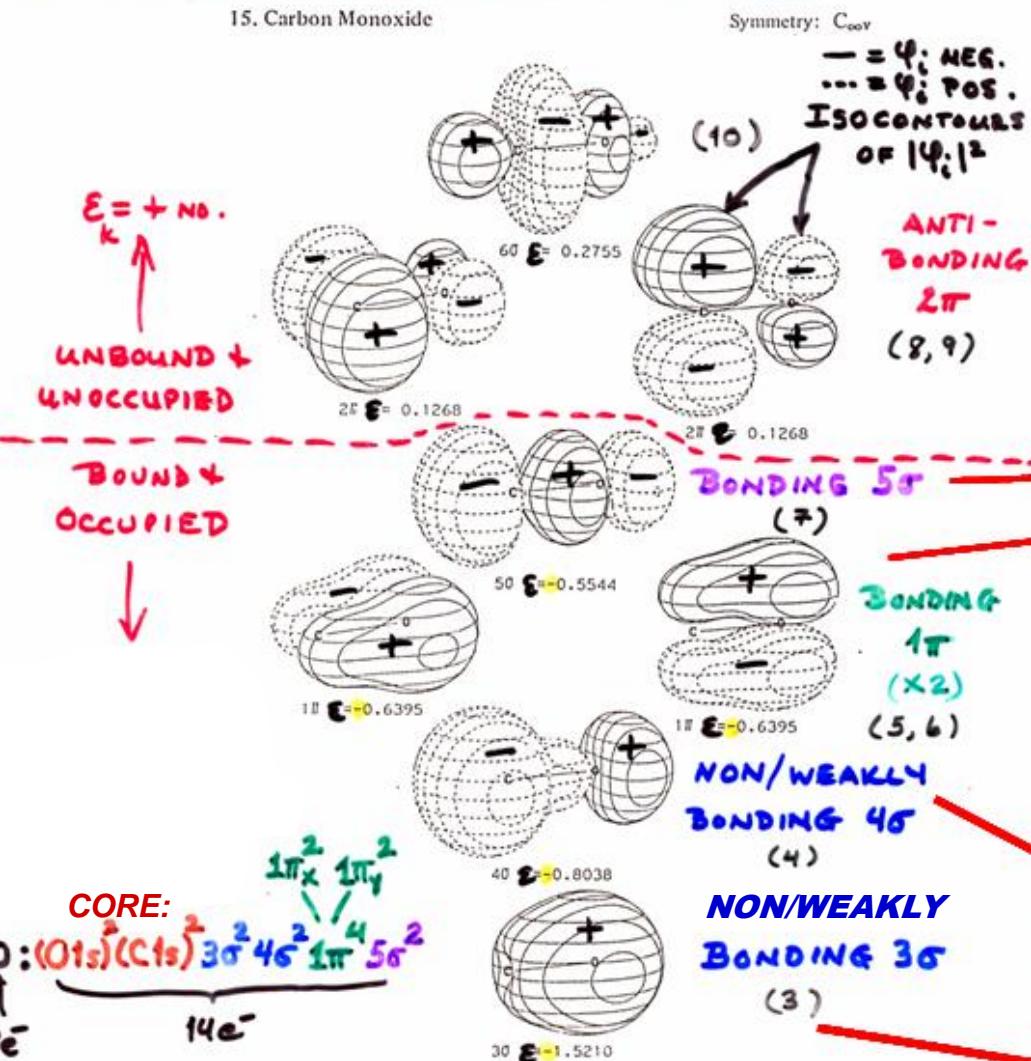


X-RAY PHOTOELECTRON SPECTRUM OF CO



The LCAO or tight-binding picture for CO:

15. Carbon Monoxide

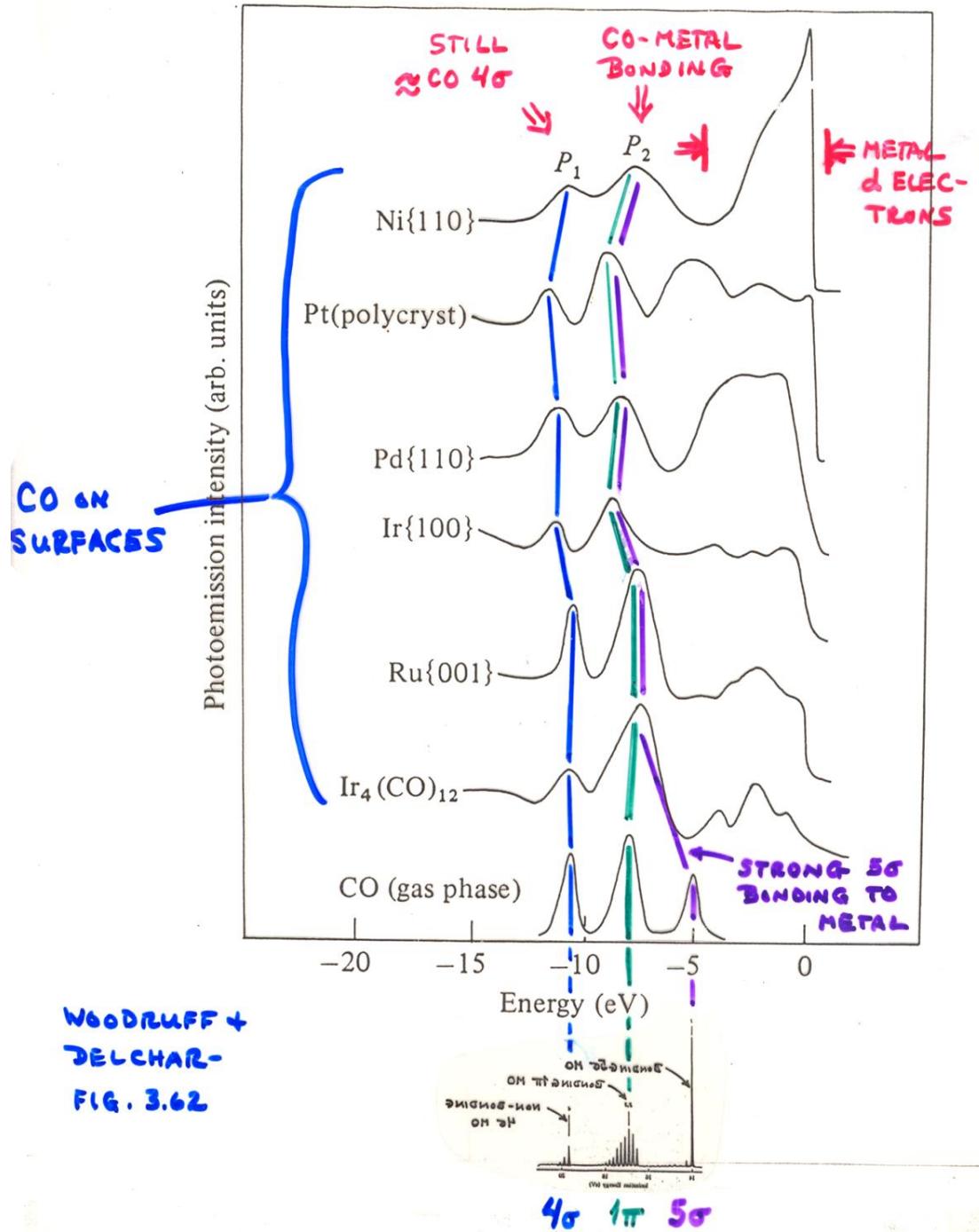


Atomic orbital makeup

$$\varphi_j^{\text{MO}}(\vec{r}) = \sum_{\text{Atoms A}} \sum_{\text{Orbitals } i} c_{Ai,j} \varphi_{Ai}^{\text{AO}}(\vec{r})$$

Atoms A
Orbitals i

Valence-level Photoelectron spectra of CO adsorbed on various transition metal surfaces



WOODRUFF +
DELCHAR-
FIG. 3.62

Theoretical Calculations of charge density for CO bound to Ni(001)- “on-top”:

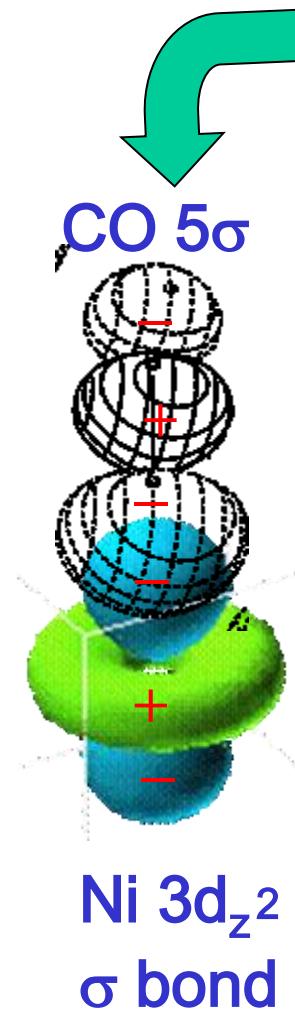
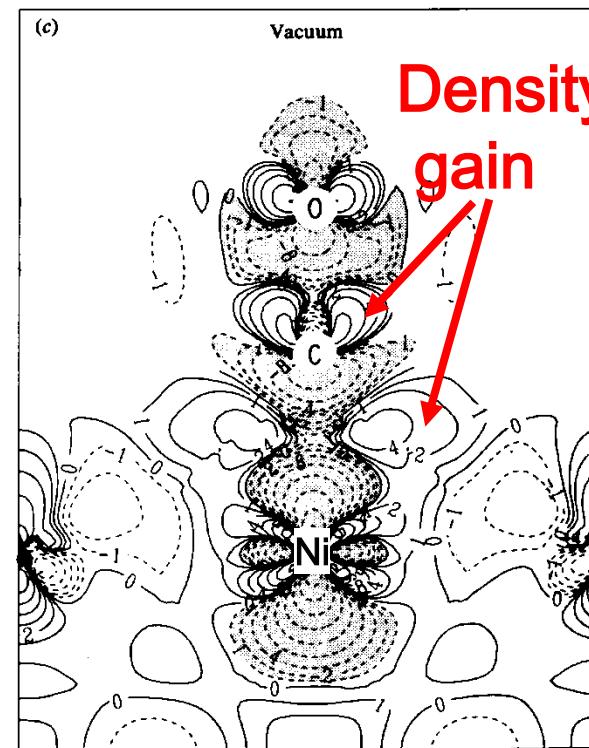
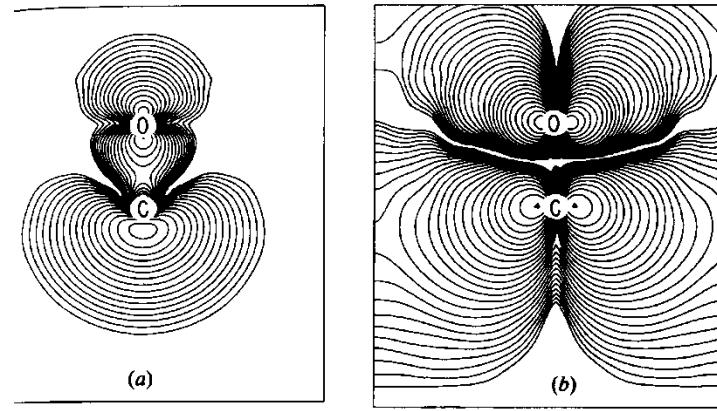
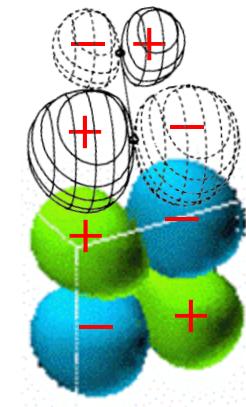


Fig. 12.14. Charge density contour plots appropriate to Ni(100) c(2 \times 2)-CO: (a) free molecule 5σ orbital; (b) free molecule 2π orbital; (c) difference between CO/Ni(100) and the superposition of clean Ni(100) and an unsupported CO monolayer. Solid (dashed) lines indicate a gain (loss) of electronic charge (Wimmer, Fu & Freeman, 1985).

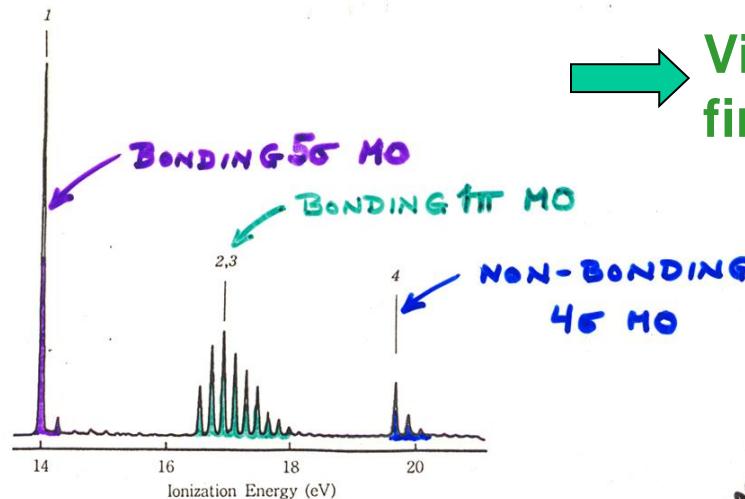


CO $2\pi = \pi^*$



Zangwill,
p. 307, plus
PRL 55, 2618 ('85)

(9) CO Carbon Monoxide

UV PHOTOELECTRON SPECTRUM OF CO

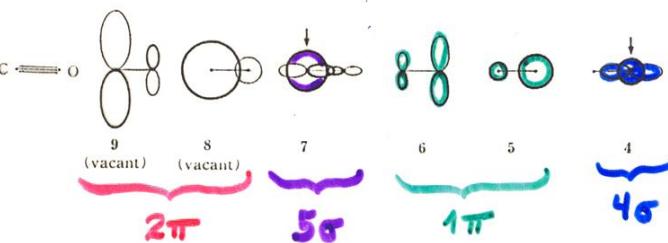
Vibrational fine structure

Expt. ^{a)} I_v (eV)	SCF MO [6-31 G] ^{b)}			CI (Ionic State) [6-31 G] ^{c)}		
	$-ε$ (eV)	MO	Character	E (eV)	State	Configuration
1 14.01	14.99	5σ (7)	$σ_{CO}$	13.11	1 ² Σ ⁺	0.93(7 ⁻¹) -0.15(6 ⁻¹ , 7 ⁻¹ , 9 ¹) _a -0.15(5 ⁻¹ , 7 ⁻¹ , 8 ¹) _a
2 16.91	17.48	1π (6, 5)	π _{bond}	16.69	1 ² Π	0.95(6 ⁻¹) ; 0.95(5 ⁻¹)
3 16.91	17.48					
4 19.72	21.69	4σ (4)	no	19.29	2 ² Σ ⁺	0.92(4 ⁻¹) +0.16(6 ⁻¹ , 7 ⁻¹ , 9 ¹) _a +0.16(5 ⁻¹ , 7 ⁻¹ , 8 ¹) _a

a) The spectrum : this work. The I_v 's : Turner *et al.* (215). See also other works : Turner and May (215 a); Carlson and Jonas (54); Gardner and Samson (104); Edqvist *et al.* (90); Potts and Williams (182 a); and Natalis *et al.* (165).

b) We used the bond length reported (A 3) ; symmetry $C_{\infty h}$. $E_{SCF} = -112.6672$ hartree. In 4-31G calculations, $E_{SCF} = -112.5524$ hartree and $-ε(eV) = 14.93, 17.41, 17.41$, and 21.60.

c) CI-II. (9, 8)=1π. $|N\rangle = 0.98$ (SCF). The results obtained in other CI levels are given in Appendix B.



Kimura et al.,
"Handbook of Hel
Photoelectron Spectra"
47

INTENSITIES IN PHOTOELECTRON SPECTRA:

- GENERAL: FINAL STATE K (K-SUBHELL + ALL OTHER DESIG.)

$$\text{INT.}_K \propto |\hat{e} \cdot \langle \Psi_{\text{tot}}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle|^2 \quad (\text{DIPOLE APPROX.})$$

- BORN-OPPENHEIMER: e⁻'S FAST, VIBRATIONS SLOW

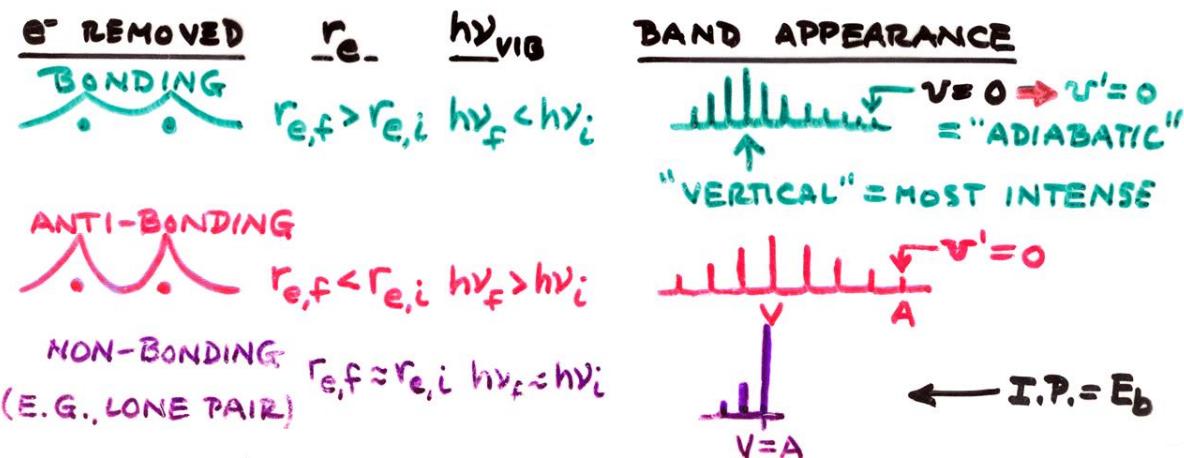
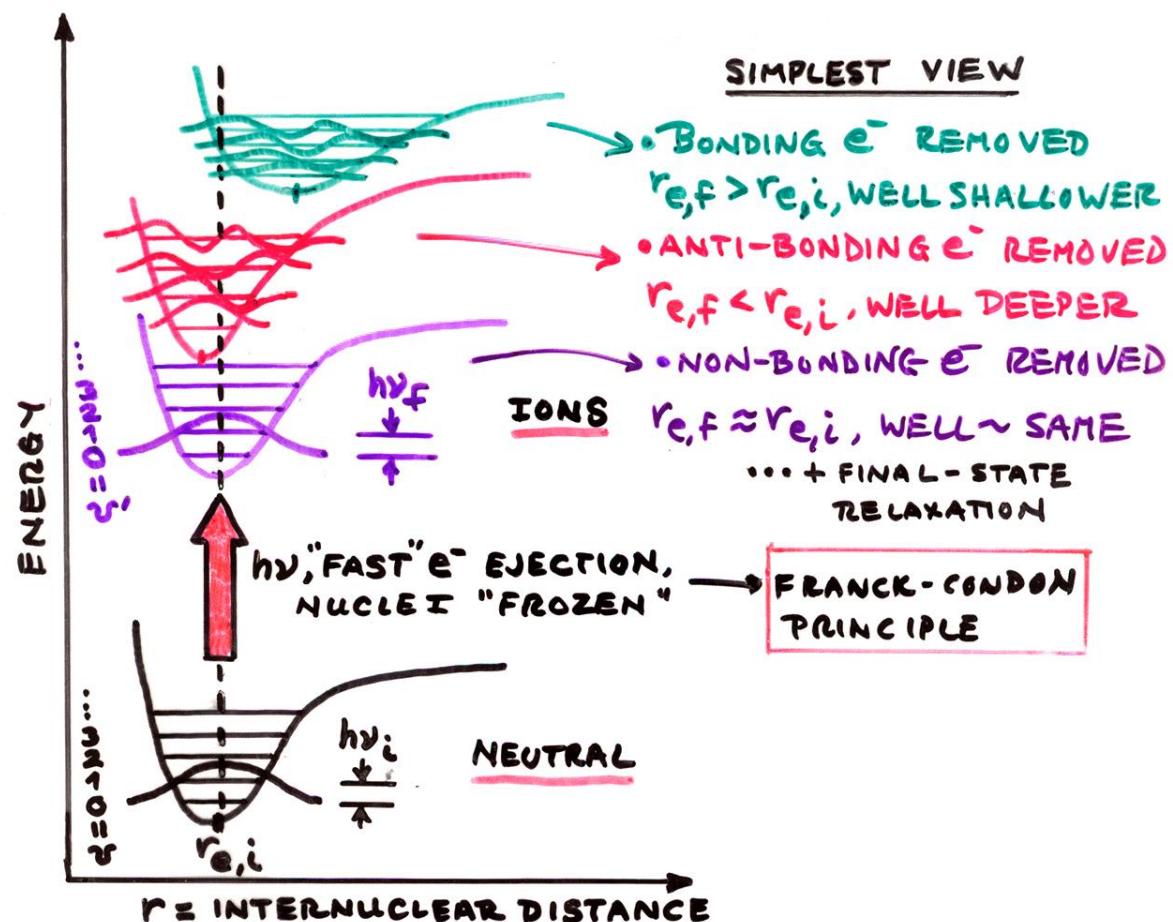
$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v.}^f | \Psi_{\text{VB}, v.}^i \rangle|^2 |\hat{e} \cdot \langle \Psi_e^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle|^2$$

FRANCK-CONDON FACTOR

VIBRATIONAL STRUCTURE IN VALENCE-LEVEL (MO) SPECTRA

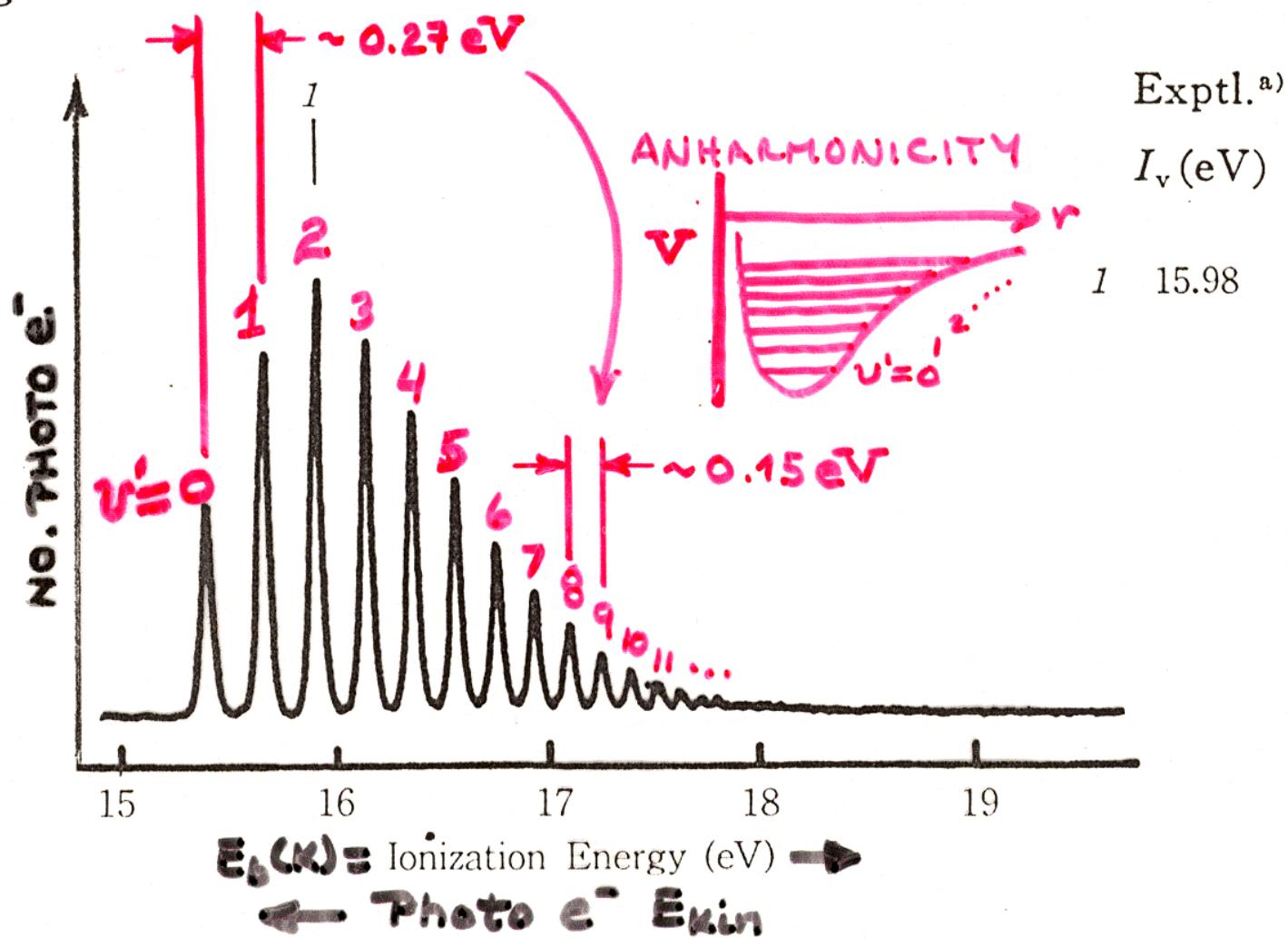
Diatom A-B example

(Also applies to core-level emission if equilibrium distance changes on forming core hole)



VIBRATIONAL STRUCTURE IN VALENCE-LEVEL (MO) SPECTRA

H_2 Hydrogen



INTENSITIES IN PHOTOELECTRON SPECTRA:

- GENERAL: FINAL STATE K (K-SUBSHELL + ALL OTHER DESIG.)

$$\text{INT.}_K \propto |\hat{e} \cdot \langle \Psi_{\text{tot}}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle|^2 \quad (\text{DIPOLE APPROX.})$$

- BORN-OPPENHEIMER: e-'s FAST, VIBRATIONS SLOW

$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\hat{e} \cdot \langle \Psi_e^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle|^2$$

FRANCK-CONDON FACTOR

- SUDDEN APPROXIMATION: $\Psi_K \rightarrow \Psi_f = \text{PHOTOE}^-$ (FAST)



$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\underbrace{\langle \Psi_e^f(N-1, K) | \Psi_e^{i-}(N-1, K) \rangle}_{|\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_k \rangle|^2} |^2$$

SAME SUBSHELL COUPLING +
 $\hookrightarrow \text{NORMAL } \frac{dG_K}{d\Omega}$ TOTAL L,S → "MONOPOLE"

- SLATER DETS. FOR $\Psi_e^f = \det(\Psi'_1 \Psi'_2 \dots \Psi'_{K-1} \Psi'_{K+1} \dots \Psi'_N)$

$$\Psi_e = \det(\Psi_1 \Psi_2 \dots \Psi_{K-1} \Psi_{K+1} \dots \Psi_N)$$

$$\text{INT.}_K \propto |\langle \Psi_{\text{VB}, v}^f | \Psi_{\text{VB}, v}^i \rangle|^2 |\langle \Psi'_1 | \Psi'_1 \rangle|^2 |\langle \Psi'_2 | \Psi'_2 \rangle|^2 \dots$$

$$|\langle \Psi'_{K-1} | \Psi'_{K-1} \rangle|^2 / |\langle \Psi'_{K+1} | \Psi'_{K+1} \rangle|^2 \dots |\langle \Psi'_N | \Psi'_N \rangle|^2$$

spin-orbit +

$$|\hat{e} \cdot \langle \Psi_f | \vec{r} | \Psi_e \rangle|^2$$

1e- DIPOLE → $d\sigma/d\Omega$

(N-1)e- SHAKE-UP/
 SHAKE-OFF →
 "MONOPOLE"

"Basic Concepts of XPS"
 Chapter 3.D.

- PLUS DIFFRACTION EFFECTS IN Ψ_f ESCAPE