Reading and Problem Assignments for Physics 243A Surface Physics of Materials: Spectroscopy, Fall, 2014

READING:

WOODRUFF AND DELCHAR, "MODERN TECHNIQUES OF SURFACE SCIENCE", 2 ND EDITION
Chapter 1 Chapter 2: Sections 2.1, pp. 22 (bottom) 22 (for) on Wood natation for ourford structures
Chapter 2: Sections 2.1, pp.22 (bottom)-23(bp) on wood notation for surface structures,
2.4, and 2.5 (pp. 51-57), 2.5.0 on standing waves
Chapter 3: Sections 31 32 33 35
 ZANGWILL "PHYSICS AT SURFACES" DOWNLOADABLE CHAPTERS 1-5 (SEE COURSE WEBSITE)
Chapter 1: Everything except "The roughening transition"
Chapter 3: pp. 28-34, pp. 49-52 on STM. Pages 85-8, 192-196, 204-212
Chapter 2: All
Chapter 4: Introduction, with lighter reading of The jellium model, One-dimensional band
theory, and Three-dimensional band theory, and detailed reading of Photoelectron
spectroscopy, Metals, and Alloys
Chapter 10: pp. 244-249 on bond lengths, no ion scattering
IBACH, "PHYSICS OF SURFACES AND INTERFACES", DOWNLOADABLE BOOK (SEE COURSE WEBSITE)
Chapter 2: 2.1, 2.2
DESJONQUERES AND SPANJAARD, "CONCEPTS OF SURFACE PHYSICS", EXCERPTS DOWNLOADABLE FROM
COURSE WEBSITE:
On equilibrium shapes of surfaces, thermodynamics, kinetics and adsorption isotherms, and on STM current calculation. No need to follo
every step, but this fills in the line of arguments in Zangwill and lecture
FADLEY, "BASIC CONCEPTS OF XPS", HANDED OUT, BUT ALSO DOWNLOADEABLE—
Sections I, II, and III, with remaining sections by the end of the course
• FADLEY, "THE STUDY OF SURFACE STRUCTURES BY PHOTOELECTRON DIFFRACTION AND AUGER ELECTRON DIFFRACTION",
PAGES 421-450 only, DOWNLOADABLE FROM COURSE WEBSITE
with other examples and exercises using the EDAC web program introduced in lecture
• ATTWOOD, DOWNLOADEABLE EXCERPT ON SYNCHROTRON RADIATION FROM THE BOOK
"Soft X-Rays and Extreme Ultraviolet Radiation" (see course website)
• SIX READING DOWNLOADS FROM THE COURSE WEBSITE:
1) Molecular orbital basics
2) Tight-Diffunity Dasics 3) Delengueres and Spanjaard handout Photoelectron Diffraction Theory/Debue Waller Factors
a) Dependences and Spanjaard nandout-motorelection Dimaction meory Debye-water Factors
5) Brief Manual for SESS spectral simulation program
6) Brief Manual for CTM4XAS20 charge-transfer multiplet simulation program
[7] Optional only for physics students: Basic theory for the Hubbard Model of bonding]
r., - F., - F.,

PROBLEM SET 3-FINAL: Not all problems assigned

Problems 3.3, 3.4, 4.1, 4.2, 4.4, 4.5, 5.1, 5.2, 5.3, 5.4, 5.7, 5.8, 5.9(a),(b) only, 5.10, DUE FRIDAY, 12 DECEMBER

REMAINING LECTURE SCHEDULE:

NO LECTURES: 2 DECEMBER AND 9 DECEMBER 12:10 MAKEUP + 2:10 REGULAR LECTURES: 25 NOVEMBER, 4 DECEMBER, AND 11 DECEMBER

TOUR OF LBNL: SATURDAY, 13 DECEMBER

FINAL EXAMINATION: TUESDAY, 16 DECEMBER, 10:30 AM-12:30 PM, PHYSICS 285 (unless moved by unanimous consent)



Outline—Here to end of quarter

- •Core-level chemical shifts: Koopmans', relaxation, the potential model
- •Various other final state effects providing information in corelevel spectra
- •Valence-band spectra: low-energy UPS limit and high-energy XPS limit
- •Photoelectron diffraction, extended x-ray absorption fine structure (EXAFS, XAFS)
- •Photoelectron spectroscopy at realistic pressures in the multi-Torr range
- •Photoelectron microscopy: adding lateral spatial resolution in 2 dimensions





CHEMICAL SHIFTS IN ADSORBATE & SUBSTRATE





> RELAXATION, SCREENING, CONFIGURATION INTELACTION, SELF-ENERGY EFFECT ALWAYS PRESENT; ANDERSON IMPURITY MODEL ETC.



Figure 18 -- Plot of carbon ls binding energies calculated via Koopmans' Theorem against experimental binding energies for several carbon-containing gaseous molecules. For some molecules, more than one calculated value is presented. The slope of the straight line is unity. The two scales are shifted with respect to one another by 15 eV, largely due to relaxation effects. All of the theoretical calculations were of roughly double-zeta accuracy or better. (From Shirley, reference 7.)

> "Basic Concepts of XPS" Figure 18





FREE-ION (INTRAATOMIC) ASPECTS OF SHIFTS: KOOPMANS' THEOREM & CLASSICAL CHARGED SHELL

"Basic Concepts of XPS" Figure 19 TEROVAL ADDITION OF VALENCE C CHARGE IN BONDING SHIFTS ALL INNER C Eb'S = EK'S BY-SAME AROUNT



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CORE-LEVEL MULTIPLET SPLITTINGS IN Mn COMPOUNDS



1

"Basic Concepts of XPS" Figure 31









Spin <u>externally</u> referenced to $\vec{k}_{h\nu}$ and \vec{M} of sample







INTENSITIES IN PHOTOBLECTRON SPEETRA:





. BORN-OFFENHEIMER : ET'S FAST, VIBRATIONS SLOW





P.R.L. 61, 2592 ('88)



Expt.--Kowalczyk et al., Phys. Rev. B<u>11</u>, 1721 (1975) Theory--Gupta and Sen, Phys. Rev. B<u>10</u>, 71 (1974) Park et al., Phys. Rev. B<u>37</u>, 10867(1988)

FILLING OF THE MOLECULAR ELECTRONIC STATES OF DIATOMIC NO AND O₂



FILLING OF THE MOLECULAR ELECTRONIC STATES OF HOMONUCLEAR DIATOMIC MOLECULES & NO



Multiplet splitting $\propto K_{1s,\pi 2p^*}$







INTENSITIES IN PHOTOBLECTRON SPEETRA:

- GENERAL: FINAL STATE K (K-SUBSHELL + ALL OTHER DESIG.) INT. K @ @. < Uf (N,K) | E . (1) / (N) / (D) POLE APPROX.</p>
- . BORN-OPPENHEIMER : E"'S FAST, VIBRATIONS SLOW

$$INT._{K} \ll \left| \left\langle \Psi_{MB,V}^{d}, |\Psi_{MB,V}^{d} \right\rangle \right|^{2} \left| \left\{ \hat{e} \cdot \langle \Psi_{1}^{d}(N, K) \left(\hat{E}, V_{1}^{d} |\Psi_{1}^{d} \rangle \right) \right|^{2} \right|^{2} \\ \xrightarrow{FRANCE - CONDON FACTOR} \\ \Rightarrow SUDDEN APPREKINATION: \Psi_{K} \rightarrow V_{F} = MeTB \bullet (FAST) \\ \psi_{1} \rightarrow \Psi_{1}^{d} \rightarrow V_{F} = MeTB \bullet (FAST) \\ \psi_{1} \rightarrow \Psi_{1}^{d} \rightarrow V_{1} \rightarrow V_{1}^{d} \rightarrow$$

• PLUS DIFFRACTION EFFECTS IN 4 ESCAPE



OVERALL: ~12 TO SHAKE-UP + 1690 SHAKE-OFF 2 28% OF EVENTS



Rennie et al., J. Phys. B: At. Mol. Opt. Phys. 32 (1999) 2691

Ethane-C 1s photoemission: "Conjugate shake-up" C 1s→unoccupied MO +occupied MO to free electron











Figure 8 -- Schematic illustration of a photoelectron spectrum involving shake-up and shake-off satellites. The weighted average of all binding energies yields the Koopmans' Theorem binding energy $-\varepsilon_k$ (sum rule (77)), and the sum of all intensities is proportional to a frozen-orbital cross section σ_k (sum rule (78)). The adiabatic peak corresponds to formation of the ground-state of the ion $(E_b(k)_1 \equiv E_b(K=1))$.

^{LE-UP}) = L- |< 𝒯[†](N-1,1) | 𝒯_R(N-L)> |² 2 = 15-25% FOR ATOMS/MALEC .


SATELLITES & CHARGE-TRANSFER SCREENING





Screening depends on lonicity/covalency→ satellite intensities can be used to measure interaction parameters



FIG. 1. Cu 2p photoelectron spectra of Cu dihalides. The lines leading to a final state with a ligand hole (\underline{L}) show a chemical shift.

Screening depends on lonicity/covalency→ satellite intensities and energy spacings can be used to measure interaction parameters



FIG. 1. Theoretical 2p core-level XPS spectra (solid line) compared with experimental data (dots) after background subtraction for Mn cations with varying valence. Emission due to the Mn *LMV* Auger peak is observed on the high-bindingenergy side of the $2p_{1/2}$ spin-orbit peak, partially obscuring the $2p_{1/2}$ satellite structure.



The Hubbard model-mixing a localized on-site picture and a delocalized band picture



See problem 5 in Chapter 32 of Ashcroft and Mermin, "Solid State Physics", which this paper goes through in nice detail:

B Alvarez-Fernández and J A Blanco

Eur. J. Phys. 23 (2002) 11–16 +243A download at website

 $H = h_1 + h_2 + V_{12}$ (1) where h_1 and h_2 are one-electron Hamiltonians and V_{12}^{-1} is the Coulomb repulsion potential between the two electrons when they are found to be on the same atom. In order to solve the problem we shall use the following procedure. First we consider a hydrogen molecule in which an atom at \vec{R} is described in the spatial representation by a single orbital electronic level $|\vec{R}\rangle$. When there is no electron on the atom $|\vec{R}, 0\rangle_{\text{vacuum}}$, i.e. an empty level, the energy is zero, if there is one electron of either spin in the level $|\vec{R}, \uparrow\rangle_{up}$ or $|\vec{R}, \downarrow\rangle_{down}$ its energy is E_0 , and if there are two electrons of opposite spins in the level $|\vec{R}, \uparrow \downarrow \rangle_{\text{singlet}}$ the energy is $2E_0 + U$. The last additional positive energy U represents the intra-atomic Coulomb repulsion between the two localized electrons = $J_{1s,1s}$. The amplitude for tunnelling is represented by the off-diagonal term in the one-electron Hamiltonian bonding

$$\langle \vec{R}|h|\vec{R}'\rangle = \langle \vec{R}'|h|\vec{R}\rangle = -t \tag{2}$$

The same U and t used in the de Groot multiplet program (to come later) $\sqrt{1}$

The Hubbard Model (continued)

Ground-state Hubbard energy:

Ground-state Hubbard energy:

$$E_{\text{Hubbard}} = 2E_0 + \frac{1}{2}U - \sqrt{4t^2 + \frac{1}{4}}U^2.$$
And wave function:

$$\Phi_{\text{Hubbard}} = \frac{1}{\sqrt{2}}\Phi_0 + \left(\sqrt{1 + \left(\frac{U}{4t}\right)^2} - \frac{U}{4t}\right)\frac{1}{2}(\Phi_1 + \Phi_2) = \frac{20.4}{-20.6}$$
With $\Phi_0 = \frac{1}{\sqrt{2}}[|\vec{R}\rangle|\vec{R}'\rangle + |\vec{R}'\rangle|\vec{R}\rangle]$

$$\Phi_1 = |\vec{R}\rangle|\vec{R}\rangle$$
Electrons on the same proton

$$\Phi_2 = |\vec{R}'\rangle|\vec{R}'\rangle$$
Electrons on the same proton

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Localized configuration interaction approach to spectrum simulation: Anderson impurity model for PS, XAS, XES



From Bocquet & Fujimori, J. Elect.Spect. & Rel. Phen. <u>82,</u> 87 (1996):

From CTM manual:

<u>C: Charge Transfer Parameters</u>

The electronic structures of transition-metal (TM) compounds, particularly the 3d TM halides, oxides and chalcogenides, have long provided intriguing problems for physicists and chemists. These compounds are highly correlated electron systems where the essential physics can be described in terms of a few interaction strengths, namely the on-site d-d coloumbic repulsion energy U, the ligand-to-metal charge-transfer energy Δ , and the ligand p-metal d hybridization strength T. Core-level X-ray photoemission spectroscopy (XPS) is a useful probe of the valence electronic structures of TM compounds, and has been successfully used in recent years to extract parame values for these interaction strengths [1-5]. In

- **Delta:** This is the charge transfer parameter $\underline{\Delta}$, which gives the energy difference between the (centers of the) $3d^{N}$ and $3d^{N+1}\underline{L}$ configurations. The effective value of $\Delta (\Delta_{eff})$ is affected by the multiplet and crystal field effects on each configuration. In the next version, the value of Δ_{eff} will be given in a parameter-output file.
- Udd: This is the value of the Hubbard U.
- Upd: This is the core hole potential. In case of XAS spectra, only the difference between U_{pd} and U_{dd} is important.
- **Hopping T:** The hopping parameters are given for the 4 symmetries in tetragonal symmetry A_1 , B_1 , E and B_2 . A_1 (z^2) and B_1 (x^2 - y^2) are part of the e_g -orbitals and E (xz, yz) and B_2 (xy) are part of the t_{2g} orbitals. In Oh symmetry the values of A_1 = B_1 and E= B_2 . (This is not yet automatic in the test-version).

Originated in the Hubbard Model: Ashcroft and Mermin, pp. 689-691

E.g.—Crystal field in Mn³⁺ & Mn²⁺ with negative octahedral ligands





FIG. 6. Fits of the cluster model results with the experimental $2p_{3/2}$ spectra of the manganese dihalides. The parameters used are listed in Table II. A Lorentzian broadening is 2.6-3.0 eV, and a Gaussian broadening of 1.2 eV (FWHM) was used.





For octahedral coord.:
$T_{\sigma} = \sqrt{3}(pd\sigma), T_{\pi} = 2(pd\pi)$
$T_{\sigma}/T_{\pi} \propto (pd\sigma)/(pd\pi) \approx -2.2$

FIG. 1. Theoretical 2p core-level XPS spectra (solid line) compared with experimental data (dots) after background subtraction for Mn cations with varying valence. Emission due to the Mn *LMV* Auger peak is observed on the high-bindingenergy side of the $2p_{1/2}$ spin-orbit peak, partially obscuring the $2p_{1/2}$ satellite structure.

Compound	d"	Valence	Δ	U	(pd \sigma)	Δ_{ef}	$U_{\rm eff}$	Main peak	Satellite peak	Ref.
SrMnO ₃ LaMnO ₃	d ³ d ⁴	4+ 3+	2.0 4.5	7.8 7.5	-1.5 -1.8	-0.2 1.8	7.1 6.8	d4 <u>L</u> d5 <u>L</u>	$d^4 \underline{L}$ $d^6 \underline{L}^2$	This work This work
MnO	d 5	2+	6.5 7.0	7.0 7.5	-1.1 -0.9	8.8	11.6	d [€] L	d ⁵	This work 3
From later paper (see next slides) 6.5 6.0 $-1.99/\sqrt{3} = -1.1$ U/Q = 0.7-1.0, a best fit no. is 0.83.								no. is 0.83.		





INTENSITIES IN PHOTOBLECTRON SPECTRA!

GENERAL: FINAL STATE K (K-SUBSHELL + ALL OTHER DESIG.) THT. C O. CHE (N. K) EF I P'(N)> (2 (DIPOLE APPROX. BORN-OPPENHEINER : "'S FAST, VIBRATIONS SLOW P. - YE = PHOTO O (FAST) · SUDDEN APPRORIMATION : INT. & I < W , & , i I W , w / 2 | < W (N-1, K) | W (N-1, K) > 2 ادَر جور المراج SAME SUBSHELL COUPLING + 4 NELMAL LIS-"MONOPOLE" · SLATER DETS. FOR Ye = dat (4'4' ... 4' 4' ... 4') INT. K K K V . (Whe .) 2/ < 4'14 >12/ < 4'14 >12. 7 $|\langle \Psi_{k-1} | \Psi_{k-1} \rangle|^2 |\langle \Psi_{k+1} | \Psi_{k+1} \rangle|^2 \cdots |\langle \Psi_{k+1} | \Psi_{k+2} \rangle|^2 \cdot \int$ (N-1)e⁻ SHAKE-UP/ 12. < 92 17 19. >12 SHAKE-OFF \rightarrow **1e- DIPOLE** \rightarrow **d** σ /**d** Ω "MONOPOLE"

PLUS DIFFRACTION EFFECTS IN 4 ESCAPE

VIBRATIONAL STRUCTURE IN VALENCE-LEVEL (MO) SPECTRA

Diatomic A-B example

(Also applies to corelevel emission if equilibrium distance changes on forming core hole)





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Basic energetics

$$hv = E_{binding}^{Vacuum} + E_{kinetic} = E_{binding}^{Fermi} + \varphi_{spectrometer} + E_{kinetic}$$

One-Electron Picture of Photoemission from a Surface



Valence-band photoemission: Angle-Resolved Photoemission (ARPES)







Expectations from simple directtransition theory + symmetry considerations in matrix elements



R.C. WHITE ET AL., PHYS. REV. B 35, 1147 (1987)



Cu: ANGLE-RESOLVED PHOTOEMISSION AND BAND-MAPPING ALONG (001)







Fe: ANGLE AND SPIN-RESOLVED SPECTRA AT Γ POINT



SrTiO₃ and La_{0.67}Sr_{0.33}MnO₃ band structures and DOS




SURFACE ELECTRONIC STATES



STRUNGLY LOCALIZED NEAR SURFACE

• BLOCH FUNCTION IN X+Y, BUT DECAYING IN Z:





ONLY EXIST WHEN NO BULK STATE EXISTS AT SAME RI = Kxî+Kyĵ; OTHERWISE MIXING OCCURS + NOT SURFACE-LOCALIBED

CONSERVATION LAWS IN VALENCE-BAND PHOTOBLECTRON SPECTROSCOPY:





Fig. 7.12. Bandstructure E(k) for copper along directions of high crystal symmetry (*right*). The experimental data were measured by various authors and were presented collectively by Courths and Hüfner [7.4]. The full lines showing the calculated energy bands and the density of states (*left*) are from [7.5]. The experimental data agree very well, not only among themselves, but also with the calculation

Fig. 4.21. Experimental dispersion of Cu(111) surface states plotted with a projection of the bulk bands: (a) Shockley state near the zone center (Kevan, 1983); (b) Tamm state near the zone boundary (Heimann, Hermanson, Miosga and Neddermeyer, 1979). Compare with Fig. 4.17.

Surface states



Fig. 4.20. Photoemission energy distribution curves from Cu(111) at different collection angles. Equation (4.32) has been used to express the electron kinetic energy in terms of the binding energy of the electron state (Kevan, 1983).





Valence-Band Photoemission at High Energy--What & Where is the "XPS Limit"?:



What all happens when you go to higher photon energies?

- non-dipole effect→the photon momentum
- angular acceptance→B.Z. averaging
- Iattice recoil, phonon creation→more
 B.Z. averaging, energy loss
-the XPS limit of full B.Z. averaging and D.O.S. sensitivity

Hussain et al., Phys. Rev. B <u>22</u>, 3750 ('80)

Valence-band photoemission—at higher energy



Hussain et al., Phys. Rev. B 34 (1986) 5226.



• Recoil leads to peak shifts and broadening: $E_{recoil}(eV) \approx \left[\frac{m_e}{M}\right] E_{kin} \approx 5.5 \times 10^{-4} \left[\frac{E_{kin}(eV)}{M(am\mu)}\right]$

Li	Ве		Table 1 Deby			bye t	e temperature and thermal conductivity"									ų.	В	С		N	0	F	-	Ne	
344 0.85	1440 2.00																0.27	223 1.2	0	2				75	
Na	Mg	9454 8															AI	Si		Р	s		CI	Ar	
1.41	400 1.56	Low temperature limit of θ , in Kelvin4Thermal conductivity at 300 K, in W cm ⁻¹ K ⁻¹ 2															428 2.37	645 1.4	8					92	
K	Ca	Sc	Ti	1	V C			Mn	Fe	Co		Ni	Cı	Cu		Zn G		Ge	Ge		Se		Br	Kr	
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			ſ	Th	Pa	a U		N	Np	Pu	Ar	n C	m	Bk	k (Cf	E	s F		n I	bN	No	1	.r	
				163 0.54			202	7 28 0.	06	0.07															

^aMost of the θ values were supplied by N. Pearlman; references are given the A.I.P. Handbook, 3rd ed; the thermal conductivity values are from R. W. Powell and Y. S. Touloukian, Science 181, 999 (1973).

Silicon--Debye-Waller Factors



Plucinski, et al. PRB <u>78</u>, 035108 (2008); Phys. Rev. B 84, 045433 (2011) Some classic cases in the XPS limit:





Tungsten--Debye-Waller Factors and Recoil Energies





Present if vibrations stiff enough (Debye T high enough), but suppressed as temperature is raised.

> Hussain et al., Phys. Rev. 22, 3750 (1980)

Effect of photon momentum on k conservation: W(110) at 1253.6 eV



Hussain et al., **Phys. Rev. 22**, 3750 (1980)

Tungsten--Debye-Waller Factors and Recoil Energies





ng-8, Nature Materials <u>10,</u> 759





Gray, Minar, Dubon (sample) et al., TBP

GaAs Band Structure

Atomic-Orbital Character and Charge Density Contours



Ylvisaker and Pickett



Outline—Here to end of quarter

- •Core-level chemical shifts: Koopmans', relaxation, the potential model
- •Various other final state effects providing information in corelevel spectra
- •Valence-band spectra: low-energy UPS limit and high-energy XPS limit
- •Photoelectron diffraction, extended x-ray absorption fine structure (EXAFS, XAFS)
 - •Photoelectron spectroscopy at realistic pressures in the multi-Torr range
 - •Photoelectron microscopy: adding lateral spatial resolution in 2 dimensions